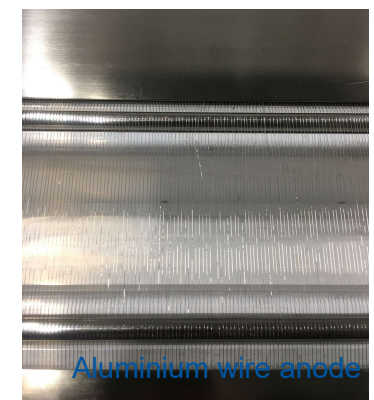
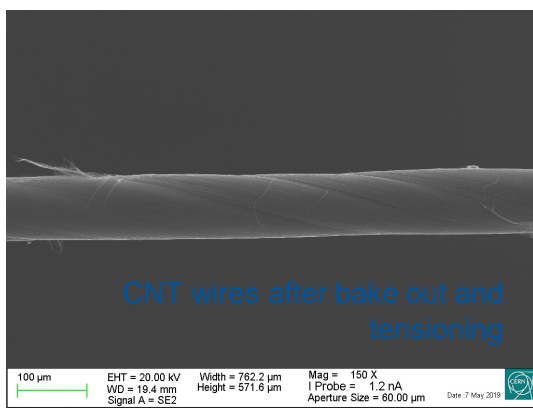




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# Status of the electrode material R&D for electrostatic septa at CERN

J. Borburgh, B. Balhan, L. Jorat, A. Prost



# Outline

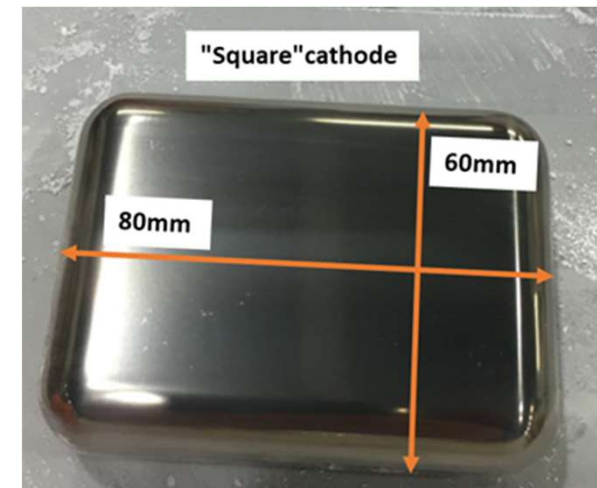
- ❖ Introduction
- ❖ HV test facility
- ❖ Anode material test results
- ❖ Cathode material test results
- ❖ Conclusion and outlook

# Introduction

- Equipment activation reduction using low density materials for septum wires and support is very promising [1].
- Anode wire options assessed.
- Cathode materials performance assessed.

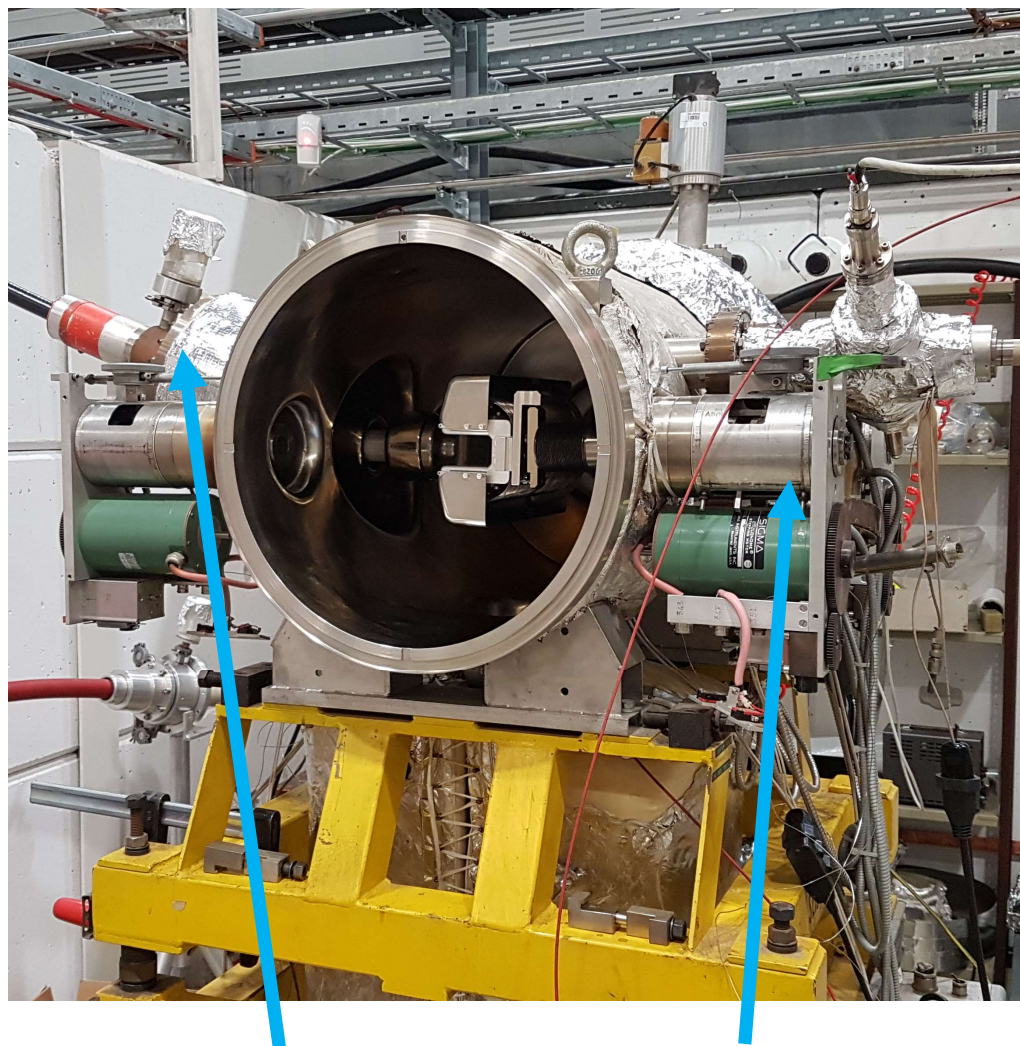
# HV test facility 1/2

- Former LEAR septum is used as a test facility.
- Adjustable cathode gap (5 - 40 mm), nominal 10 mm.
- Septum anode wire 40 x 300 mm (V x H).
- Used existing test cathode.



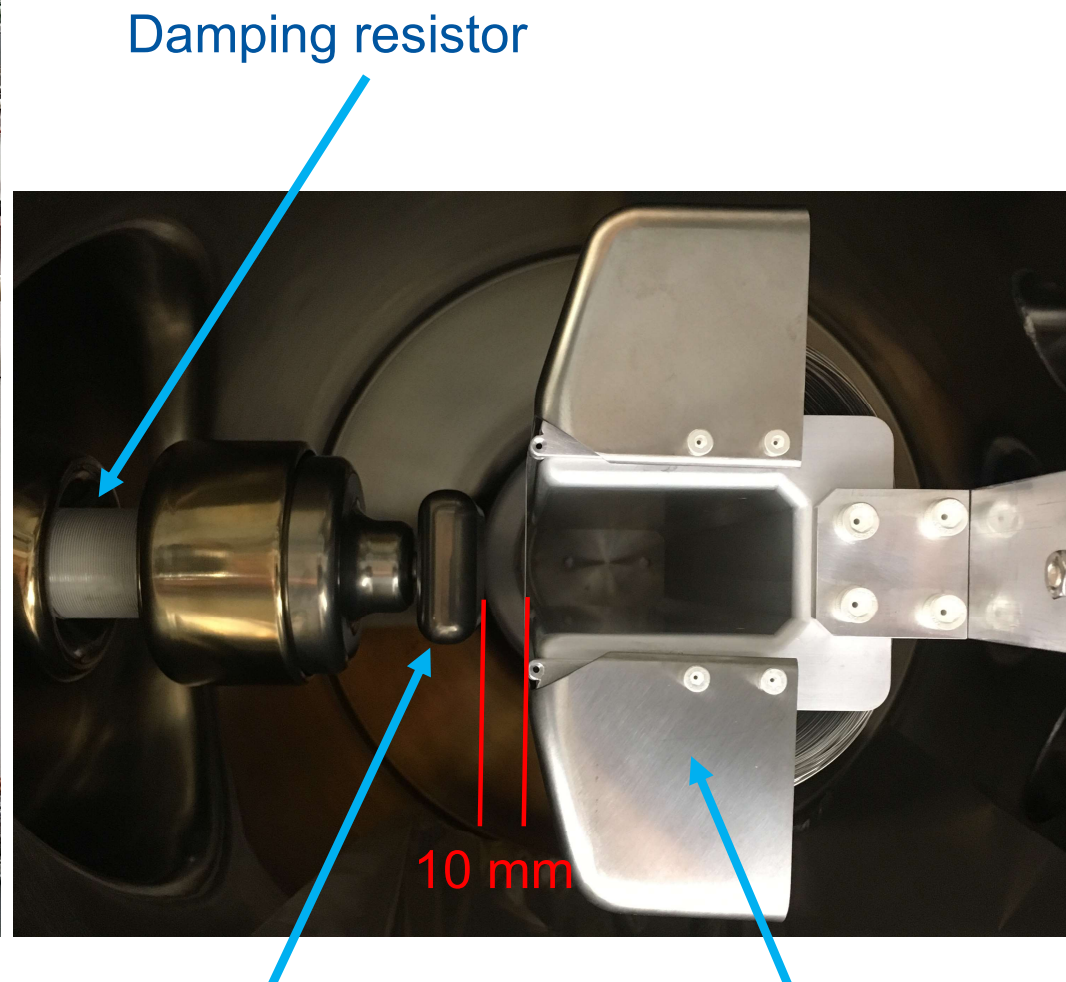


# HV test facility 2/2



HV feedthrough

motorisation



Damping resistor

Ti cathode

10 mm

Anode wire support

# Anode wire materials

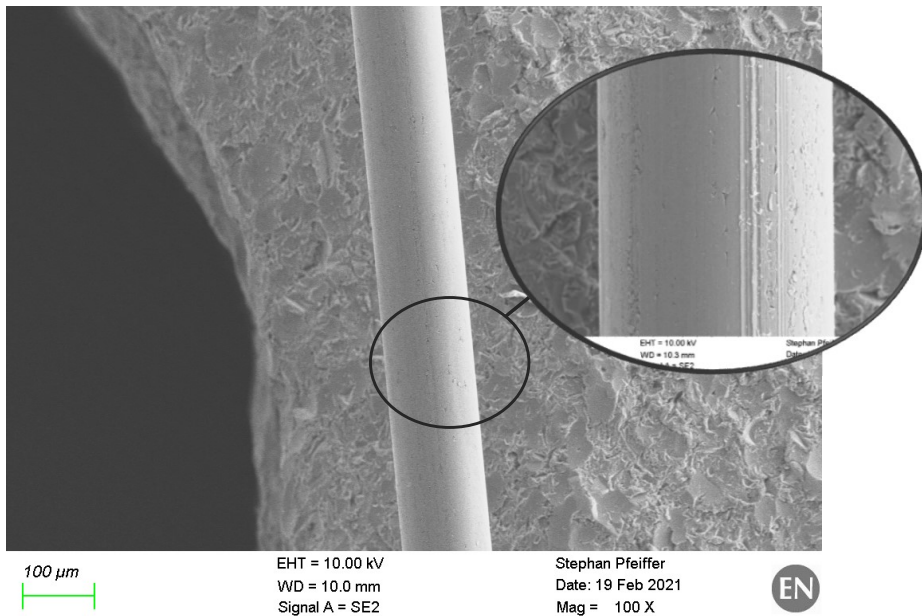
Wire material	Ø [µm]	Density [kg/l]	Alloy reference	state	Remarks
Tungsten	100	19.7	WRe (26%)	Annealed	Reference material
Stainless steel	100	7.9	316 L	Tempered hard	Not yet tested
Titanium	100	4.42	TA6V	Annealed	
Aluminium	125	2.7	AlMgSi (98/1/0.6 %)	As drawn	
Carbon Nano Tube	100	1.3	CNT wire made of 7µm CNT strands		Measured density ±0.2 kg/l

# Aluminium wire

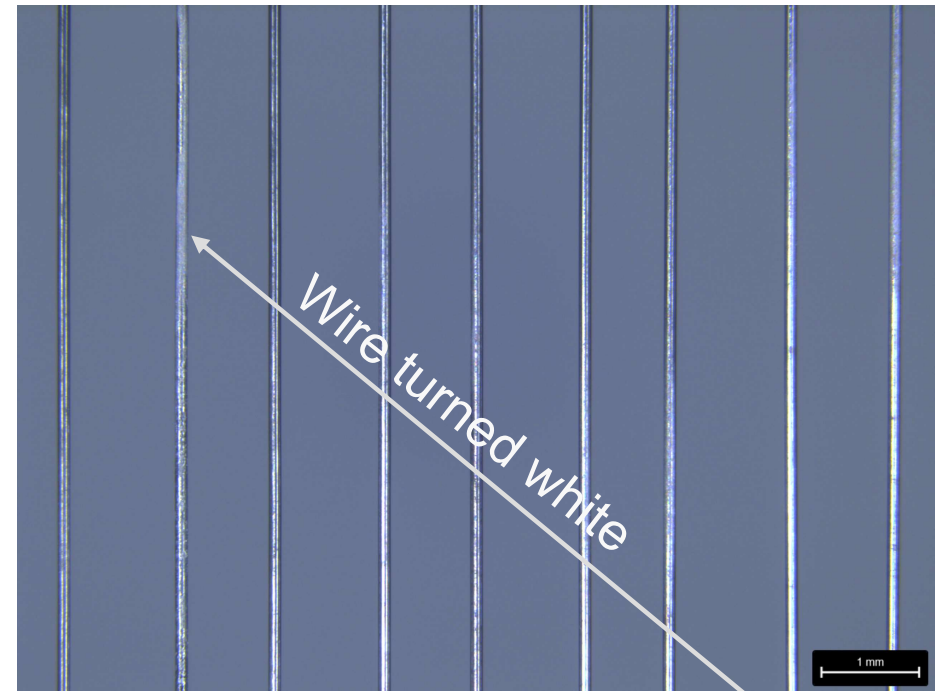
- In total 12 broken wires during conditioning, despite the initial tests with low current.
- Cathode seems to have suffered after conditioning, i.e. observed more surface irregularities than with other tests (Al deposits?, not yet determined).
- The surface of some wires that saw field are heavily deformed after conditioning.



# Aluminium wires, after conditioning 1/2

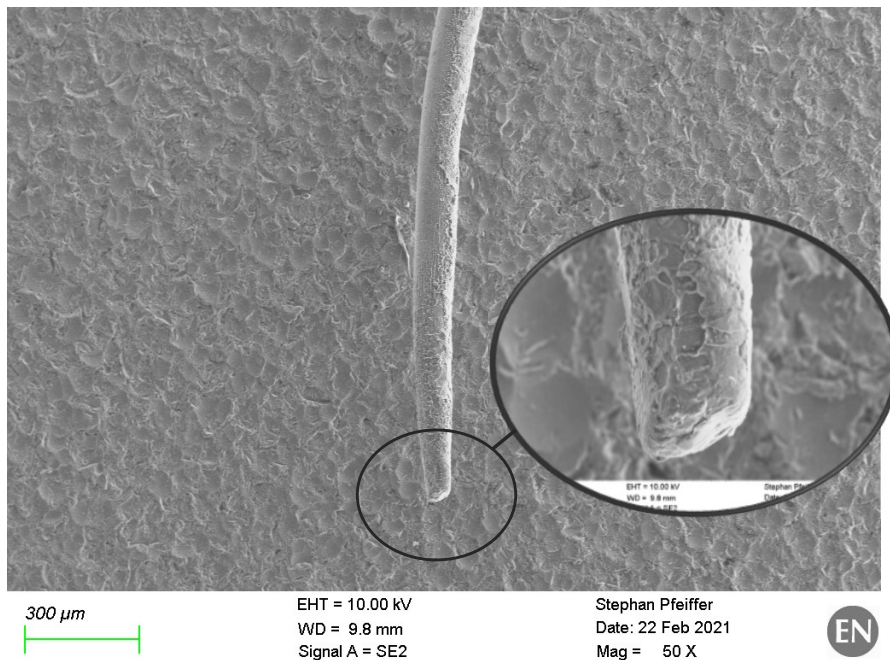


Wire surface of wire after exposure to high field; no surface damage observed.

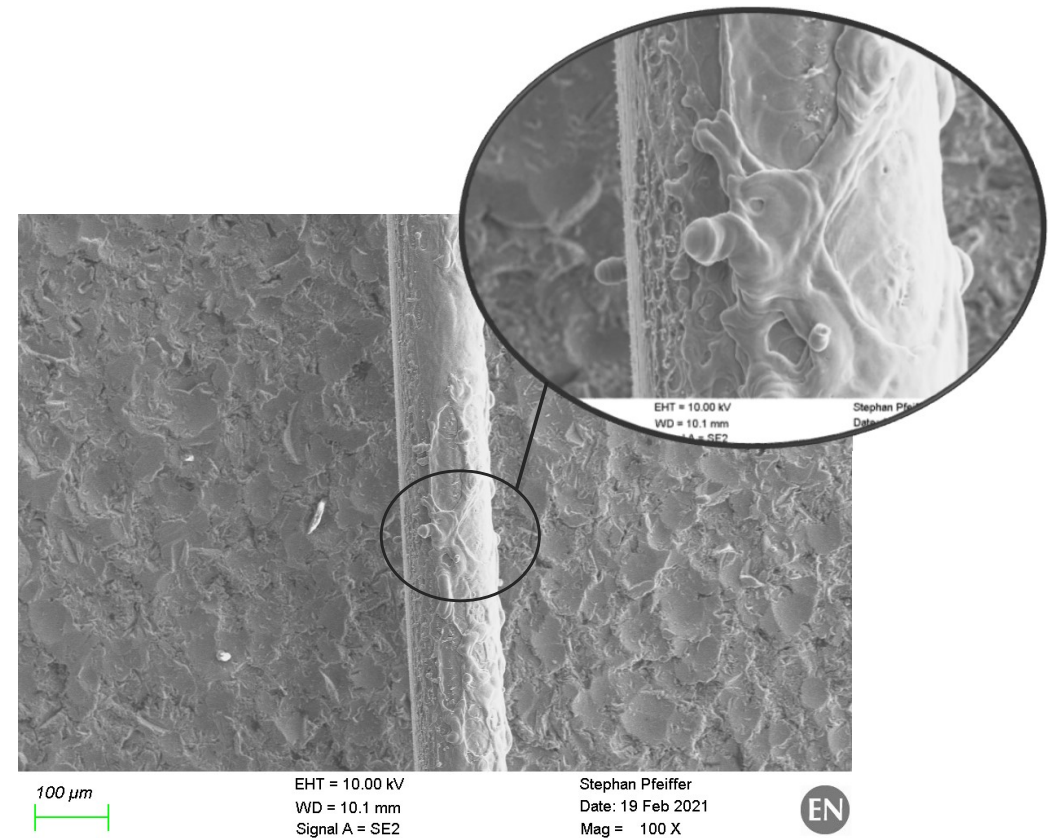


State of Al wires in the array after conditioning; some wires changed color.

# Aluminium wires, after conditioning 2/2



Extremity of a wire broken during conditioning.



‘White’ wire after exposure to high field on the right side and low field on the left. Droplets created on surface. Conditioning damage?

# Titanium wire

- The titanium revealed itself as a very springy wire: it was very hard to tension the wires straight (i.e. nicely parallel) on anode support.
- Only a few broken wires during conditioning. Mainly because higher currents were used to condition (up to 200  $\mu\text{A}$  instead of several 10s of  $\mu\text{A}$ ).
- When a **Ti wire broke (4) using a Ti cathode, welding of an end onto the cathode has occurred**, shorting the system.
- Broken Ti wires (2) didn't weld onto a stainless-steel cathode.

# CNT wire

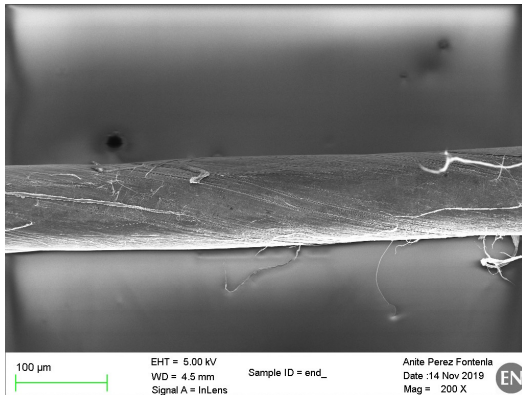
- Tested Stainless Steel and a Titanium cathode in combination with CNT wire anode.
- Similar behaviour appears independent of cathode material: 1 spark (at  $\sim 15\text{kV}$ ) provokes an irreversible current, and voltage hold off decreases to  $12\text{kV}$ .
- Could be recovered by  $\text{N}_2$  conditioning at atm. pressure up to  $15.9\text{kV}$ . Subsequently under vacuum  $20\text{ kV}$  ( $E=2\text{MV/m}$ ) could be achieved without current.

# CNT wire investigation summary

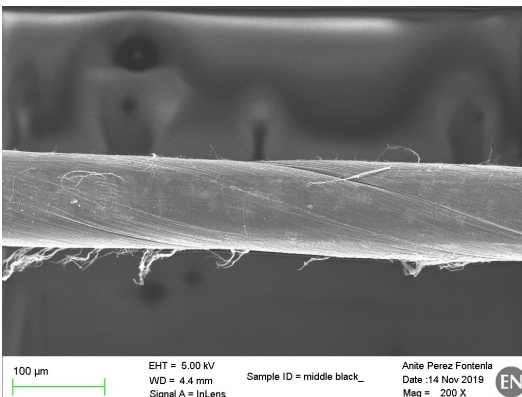
- No broken wires during conditioning (sparking): mechanically very robust.
- Once a spark occurred, the cathode seems to become an electron emitter. The current could be increased 10-fold (or more) without increasing voltage or provoking a spark. Further conditioning became impossible.
- Same behaviour independent of the cathode material used (Ti and Stainless Steel).



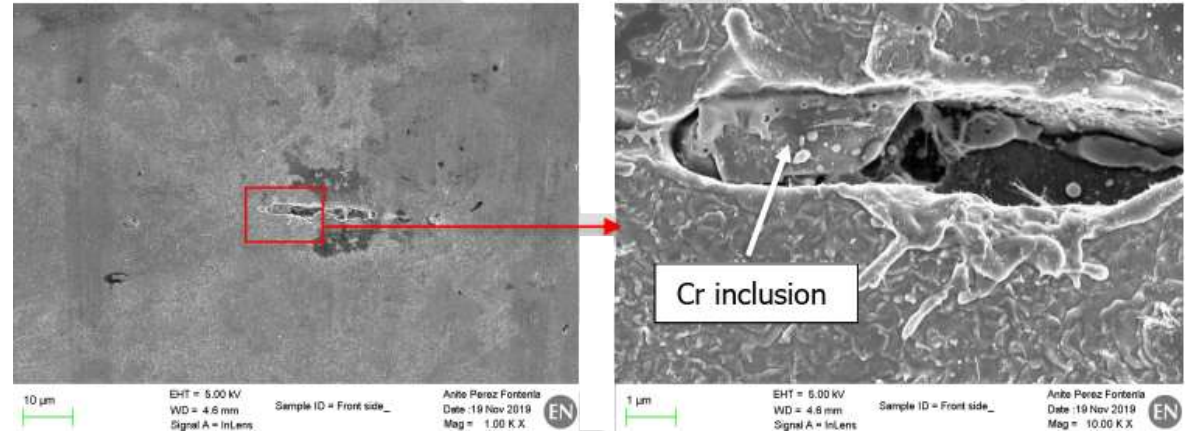
# CNT wire and dust on cathode



Zoom of a wire in low field region after conditioning. Strand damage from installation only.



Zoom of wire in high field region after conditioning. Degradation of some strands was visible.



Cathode: surface aspect at 1k magnification and detail of the molten material at the inclusion edge at 10k magnification.

**Inclusions were found on front (i.e. high field) face of the cathode, but none on the back side.**

**Hypothesis: the included CNT debris acts as electron emitter on cathode.**

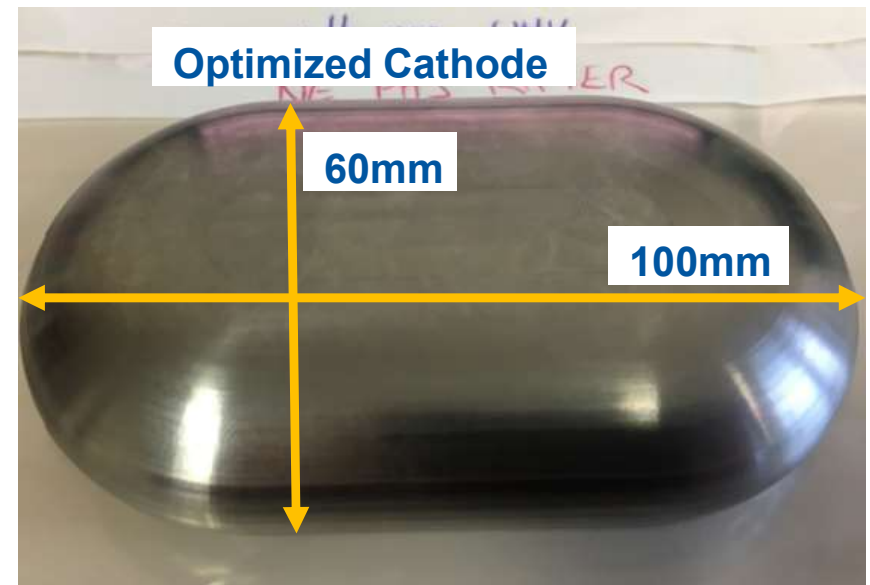
# Test summary

Cathode Material	Anode Material	Gap width [mm]	Achieved voltage [kV]	Corresponding peak field [MV/m]	Test tank #/ cathode
St. steel.	WRe	10	70	7.7	23/v4
Ti	WRe	10	130	14.3	23/v4
St. steel.	CNT	10	30	*	11/sq
Ti	CNT	10	18	*	11/sq
St. Steel	Ti	10	100	11	11/v4
Ti	Ti	10	102	11.2	11/v4
Ti	Al	10	115	12.7	11/v4
St. Steel	Al	10	115	12.7	11/sq

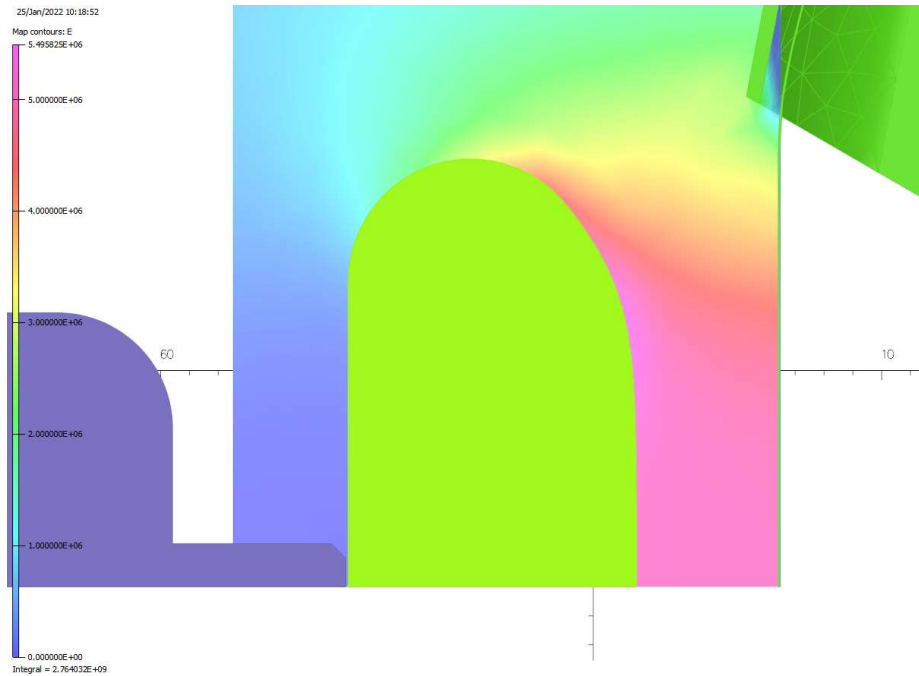
\* Field amplification factor of square cathode not available.

# Cathode material tests

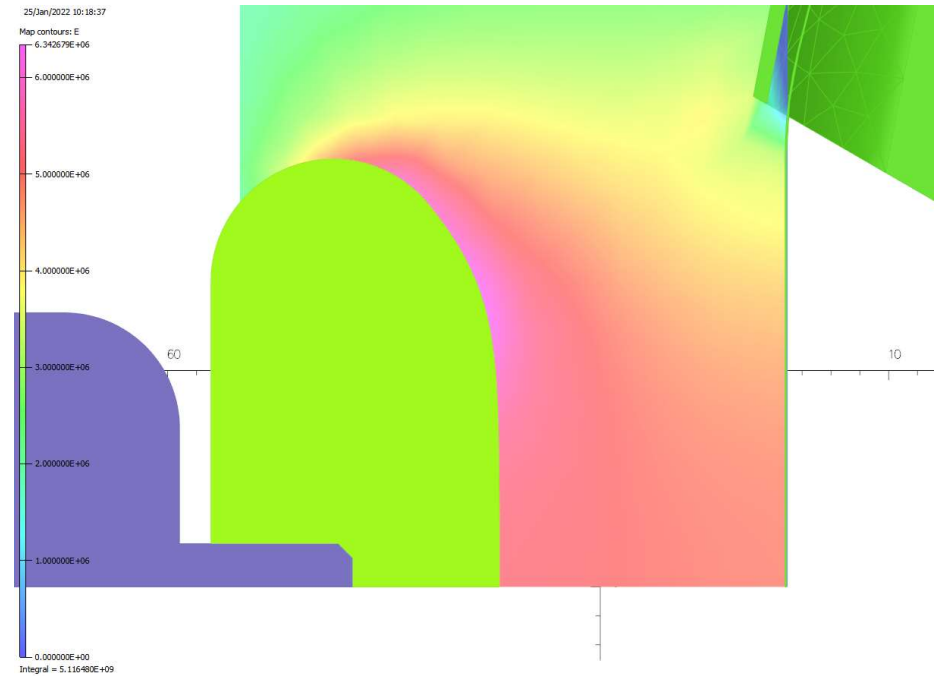
- A short cathode was designed, to minimize the field on the cathode surface.
- Manufactured from several materials to allow comparative testing.
- All tests done using a WRe wire anode.



# Cathode profile



**10 mm gap,**  
Cathode 50 kV  
 $E_{\text{gap}} = 5 \text{ MV/m}$   
 $E_{\text{max}} = 5.49 \text{ MV/m}$



**20 mm gap,**  
Cathode 100 kV  
 $E_{\text{gap}} = 5 \text{ MV/m}$   
 $E_{\text{max}} = 6.34 \text{ MV/m}$

# Aluminium cathodes

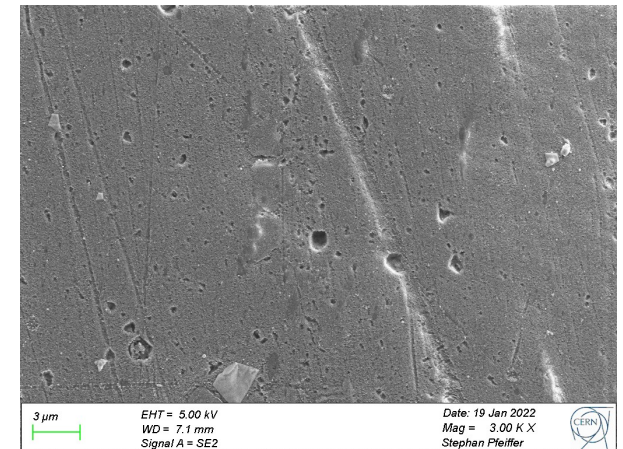
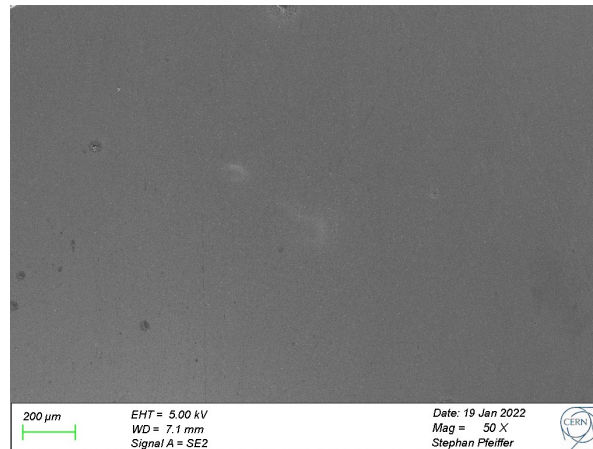
- 1) **Chromic Anodized** alloy used for small gaps  $\leq 10$  mm. Now prohibited.
- 2) **Sulfuric Anodized Peraluman 300**: used for larger gaps (used in operational CERN septa).
- 3) **Gold plated** aluminium alloy:
  - Used in the CSR ring (Max. Plack Institute, Heidelberg) to avoid  $\text{Al}_2\text{O}_3$  layer to obtain better fields (charges cannot stick to surface, altering field distribution).
  - Achieved by successive coating with various metals, and a final gold layer.
  - Test cathode plated at CERN.



# Bake-out compatibility 1/2

- Anodized Aluminium

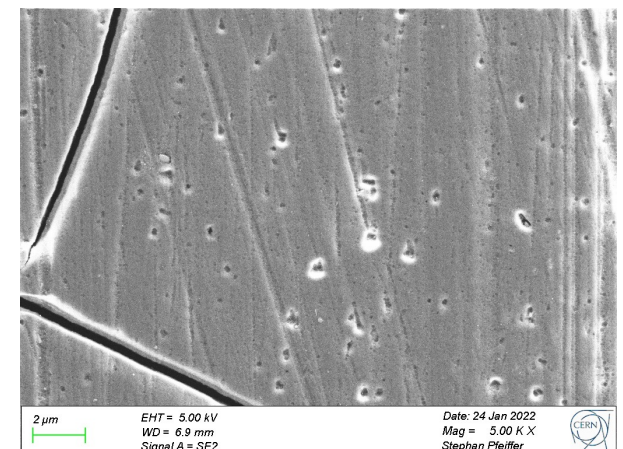
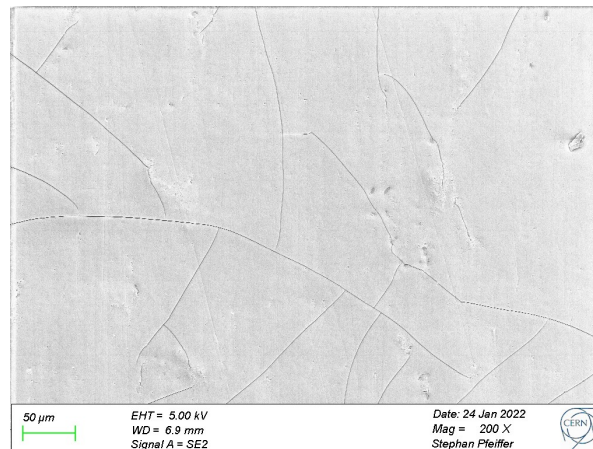
Before bake out  
(50x and 3000x)



After bake out  
(200x and 5000x)

## Conclusion:

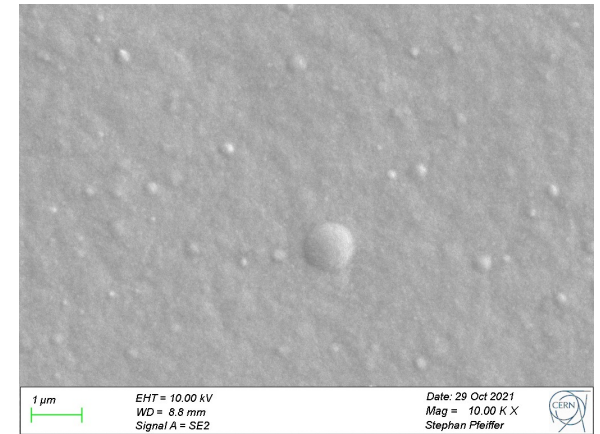
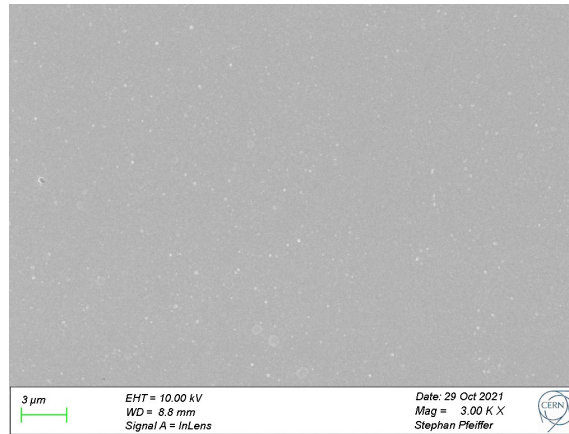
Cracks visible, HV  
performance to be verified.



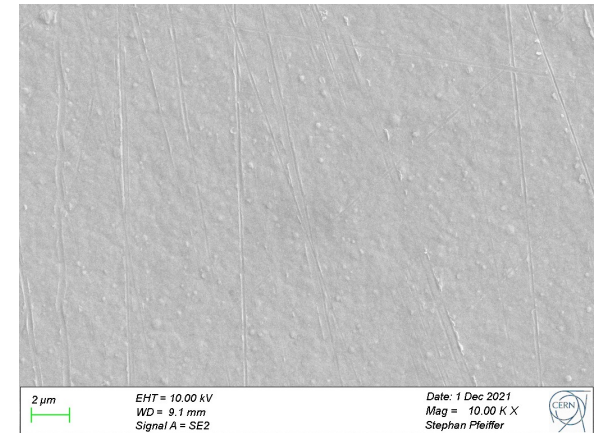
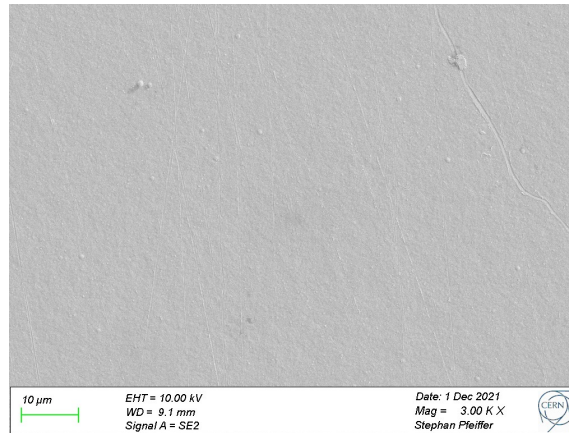
# Bake-out compatibility 2/2

- Gold-plated Aluminium cathode

Before bake out  
(3k x and 10k x)



After bake out at 200°C  
(3k x and 10k x)



## Conclusion:

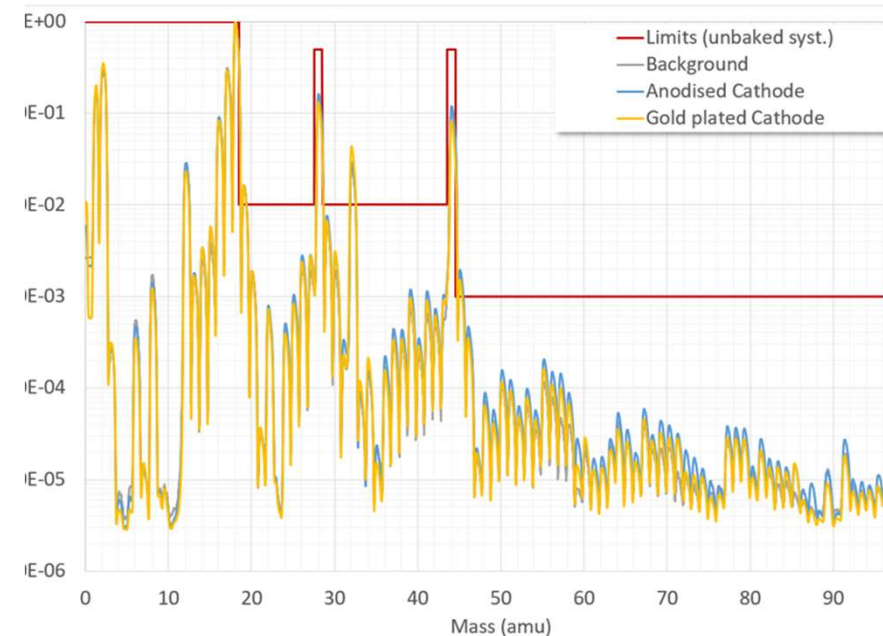
no significant change of  
surface due to bake-out.

# Vacuum performance

Cathode	Outgassing rate [mbar.l/s]	remark
Sulphuric anodized Al alloy	1.1e-5	After conditioning, no bake out
Gold-plated Al alloy	8.7e-8	After bake out, and conditioning
Stainless steel	3e-12	Typical value for reference [5]

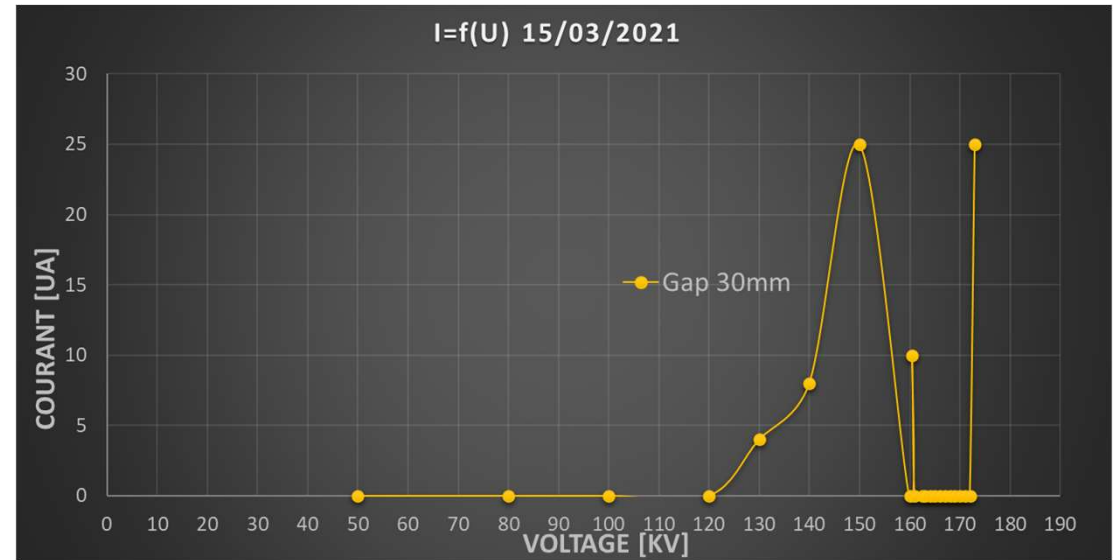
Outgassing of gold-plated Al alloy cathode much better than anodized variant.

Both cathodes residual gas analysis are compatible with CERN acceptance limits.

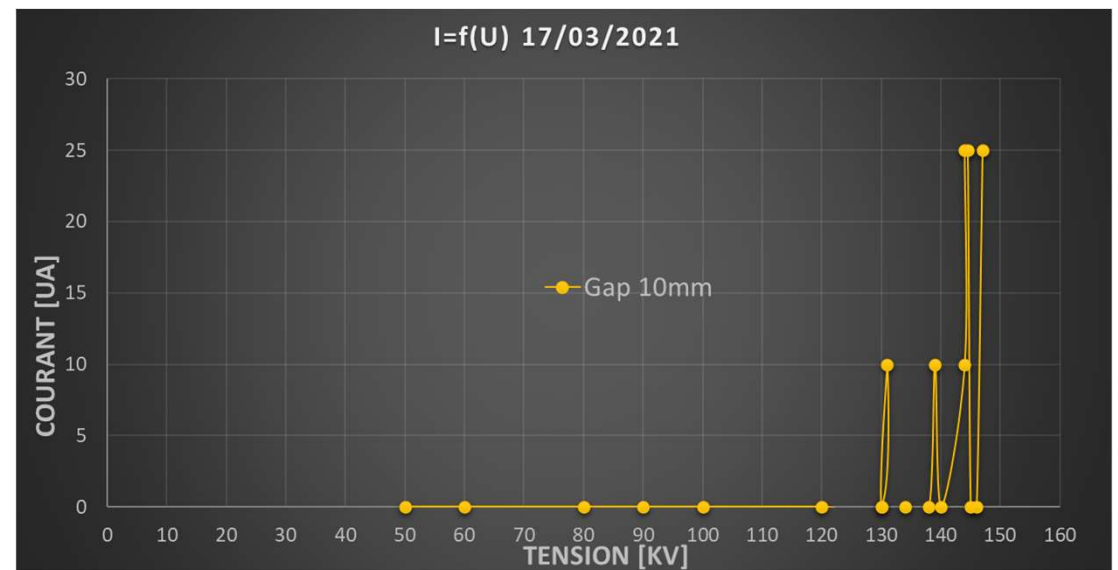


# Conditioning

- Regulated current conditioning process.
- To validate set-up, first conditioned with a 30 mm gap.
- Sulphuric anodized Aluminium cathode with WRe anode array.



10 mm gap conditioning.



## Conclusion:

Several sparks are needed to obtain stable performance.



# Max. fields achieved

Cathode Material	Anode Material	Gap width [mm]	Best voltage without current (kV)	Corresponding peak field [MV/m]
Stainless steel.	WRe	10	70	7.7
Titanium	WRe	10	130	14.3
Anodised Aluminium	WRe	10	150-160	16.5-17.6
Anodised Aluminium after bake out	WRe	10		
Gold plated Aluminium	WRe	10	130	14.3
Gold plated Al after bake out 120°C	WRe	10	130	14.3
Gold plated Al after bake out 200°C	WRe	10	130	14.3



# Conclusions and outlook

- Al and Ti anode wire materials remain of interest and warrant further study.
- Sparking with CNT anode wires led to cathode surfaces to be transformed in electron emitters.
- Gold plated aluminium cathodes permitted higher values than stainless steel and titanium, albeit still slightly less than anodised aluminium.
- Another topic of interest for operation is the spark rate. Source terms to explore:
  - temperature dependence
  - dependence of x-rays

# References

1. D. Björkman et al., “Alternative material choices to reduce activation of extraction equipment”, IPAC 2019, Melbourne Australia
2. M. Meyer, “Characterization of (CNT) carbone nano-tube and tungsten wires”, EDMS report 2142296, CERN, Geneva, 2019.
3. J. Borburgh et al., ‘Experimental Results of Low-Z Materials as a High Voltage Septum Anode’, ISDEIV2021, Padova.
4. J. Borburgh et al., ‘Low-Z anode wires testing for particle accelerator electrostatic septa’, MevArc2021
5. I. Wevers, “Degassing of different steels for the Septum”, CERN EDMS 1725086