

# Elemental Analysis with Negative Muons

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# Outline

## Introduction

Non-destructive, depth-selective, multi-elemental analysis  
Negative muon and muonic atom  
Elemental analysis method with muonic X-ray  
Muon induced nuclear reaction and gamma-ray

## Experimental

J-PARC Muon facility (MUSE)  
Experimental setup

## Examples

Chinese bronze coin and mirror, Japanese gold coin  
Extra-terrestrial sample  
Li-ion battery  
Japanese sword (lifetime method)

## Summary

# Elemental Analysis of Relics, Ancient Arts, Handiworks etc.

## Elemental composition and distribution reflect

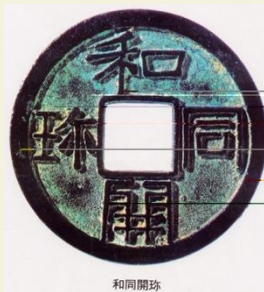
- Techniques
- Raw materials and trade
- Transition of techniques



Sampling may damage materials.  
Analysis is not practical.

# Necessity of depth-selective, non-destructive, multi-elemental analysis

Rusted or modified surface has different  
composition from the inside.



Bronze coins, bells

Plated statue

# Typical elemental analysis methods

## ICP-AES, ICP-MS

highly sensitive, quantitative,  
needs sample destruction

## XPS, PIXE

position sensitive(2D), material surface analysis,  
poorly quantitative for light elements

## Neutron activation analysis (NAA and PGA)

isotope analysis, bulk analysis, very high sensitivity for  
some nuclides, no depth-profiling ability

## SIMS

microanalysis, destructive

## Required conditions

★ Non-destructive

★ Multi-elemental

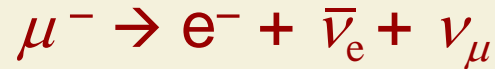
★ Depth-selective

Muonic X-ray method meets these conditions.

# What is muon?

## Elementary particle

- Charge  $-e$
- Spin:  $1/2$
- Lifetime:  $2.2 \mu\text{s}$



- No strong interaction
- Mass:  $106 \text{ MeV}/c^2$

$$\sim 207 m_e, \sim 1/9 m_p$$



## Muons in material

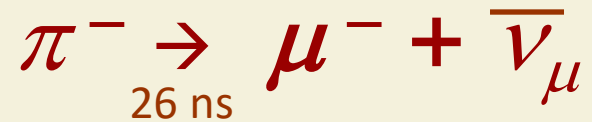
$\mu^+$ : light proton

$\mu^-$ : heavy electron

## Leptons

$e$	$\mu$	$\tau$
$\nu_e$	$\nu_\mu$	$\nu_\tau$

## Muon production

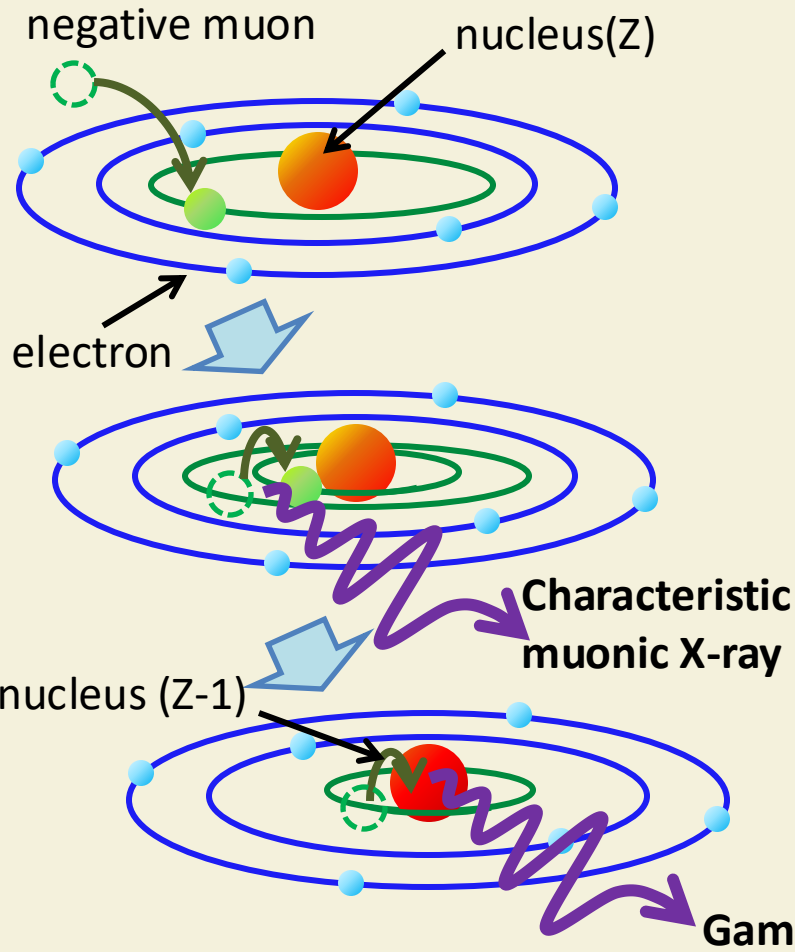


$\pi^{+-}$  are produced by

$p^+$  ( $\geq 280 \text{ MeV}$ ) + nucleus

# Negative muon and muonic atom

## Muonic atom formation and following processes

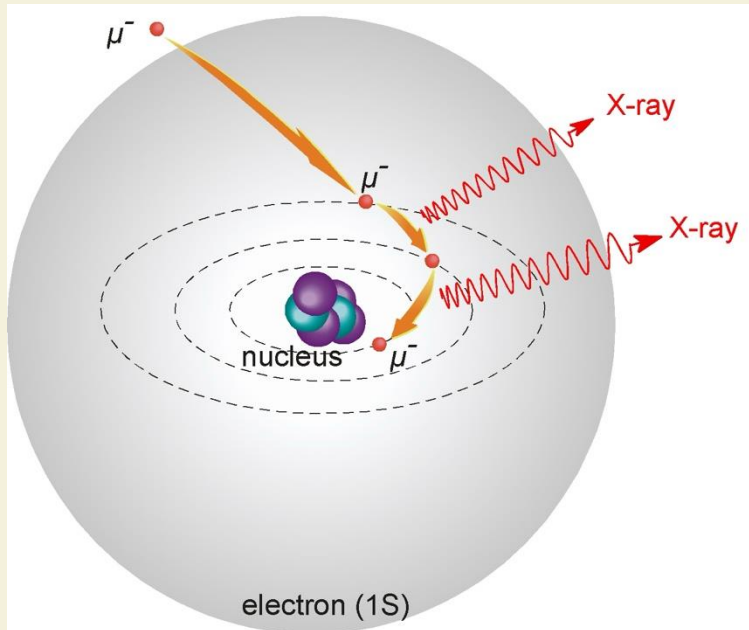


1. Muonic atom formation  
Muon capture in atomic muonic orbital
2. Muon cascading process  
**Characteristic muonic X-ray emission**
3. Muon in muonic 1s state (50-20000 ns)
4. Natural decay or muon capture in the nucleus  
**Gamma-ray emission**



# What is muonic X-ray?

Characteristic X-ray emitted during de-excitation process of muonic atom



$$E_n = -\frac{Z^2 m e^4}{8n^2 \epsilon_0^2 h^2}$$

$$r_n = -\frac{4\pi \epsilon_0 n^2 \hbar^2}{Z m e^2}$$

$$\frac{m_\mu}{m_e} \approx 207 \approx \frac{E_\mu}{E_e} \approx \frac{r_e}{r_\mu}$$

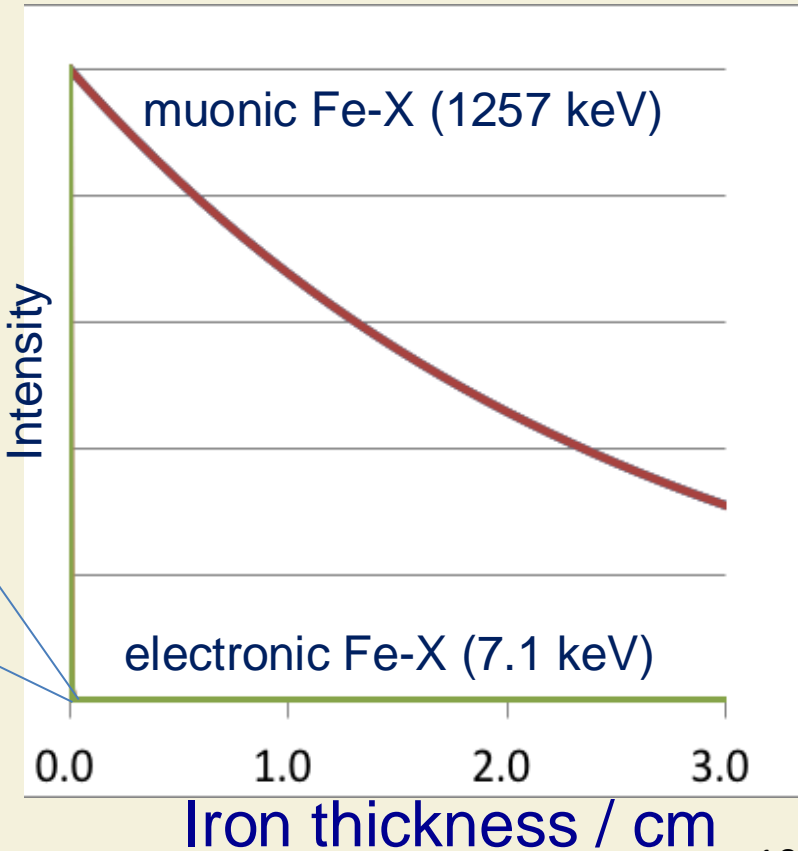
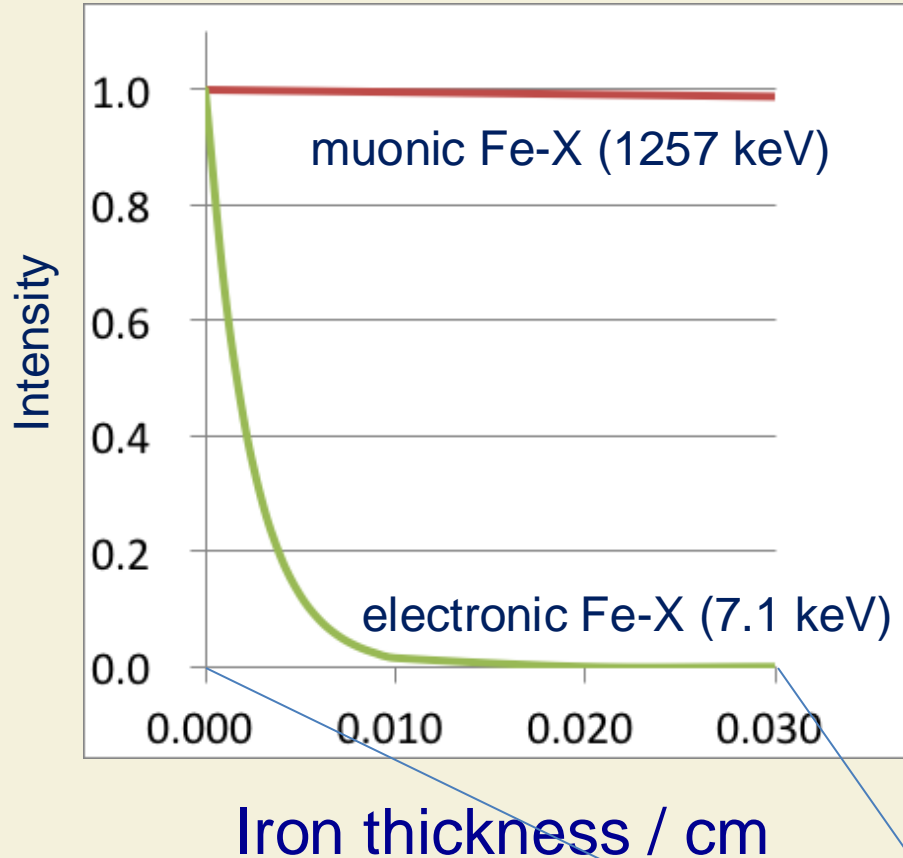
Bohr model

# Muonic X-ray Energy (keV)

Element	$K_{\alpha}$	$K_{\beta}$	$L_{\alpha}$	$L_{\beta}$
C	76	89	14	19
O	134	158	25	34
Al	347	422	66	89
Fe	1256	1704	264	357
Cu	1513	2126	307	444
Pb	5966	8466	938	1372

~200 times larger than the corresponding electronic X-ray

# Electronic and Muonic X-ray attenuation in Iron



Muonic X-ray penetrates a few cm thick Iron.

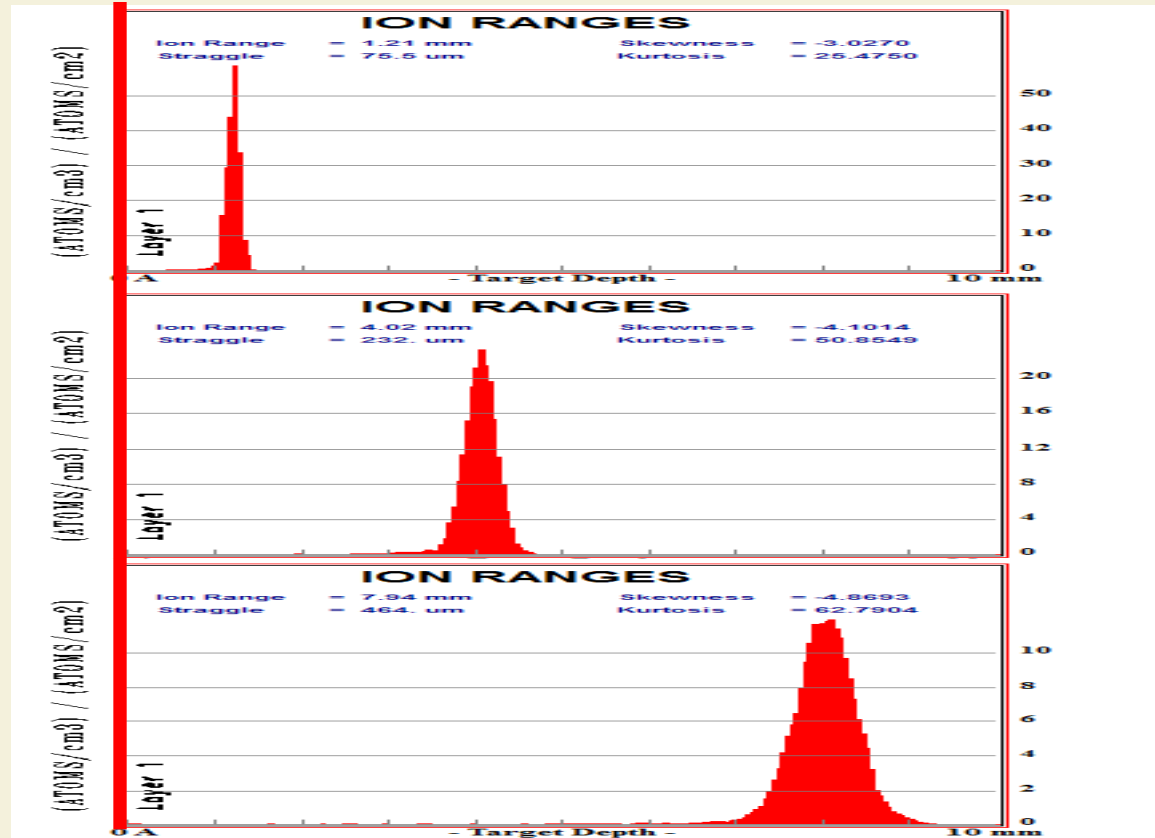
# Stopping depth of muons

Muons with a variety of momentum are produced at the same time. We choose and use muons of proper momentum depending on the depth of interest.

slow



fast



Surface

Depth



# Elemental analysis with Negative Muons

Formation of muonic atom.

Emission of characteristic muonic X-ray with specific energy to the element. **More than one muonic-Xray by one muon.**

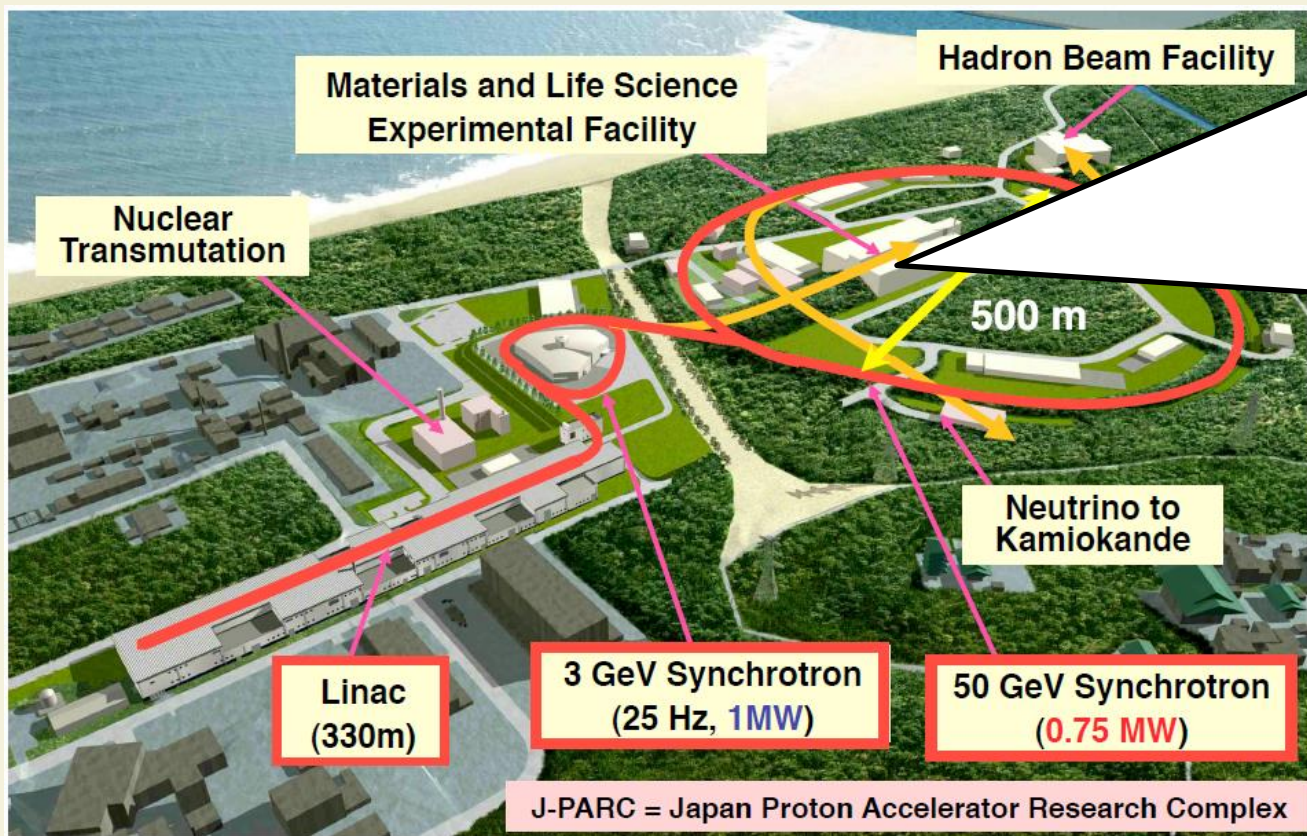
- Applicable to every element **Multi-elemental, simultaneous**  
no need of previous knowledge
- 10 keV to 10 MeV  
Observable from outside of sample **Deep inside**  
No need of vacuum -> applicable to **Huge/Porous/Bio samples**
- No chemical process  
damageless **Nondestructive**
- Depth selective by changing incident energy, easy beam scan  
**Site selective, 3D mapping**

< **Capture probability: proportional to Z with small chemical effects** >

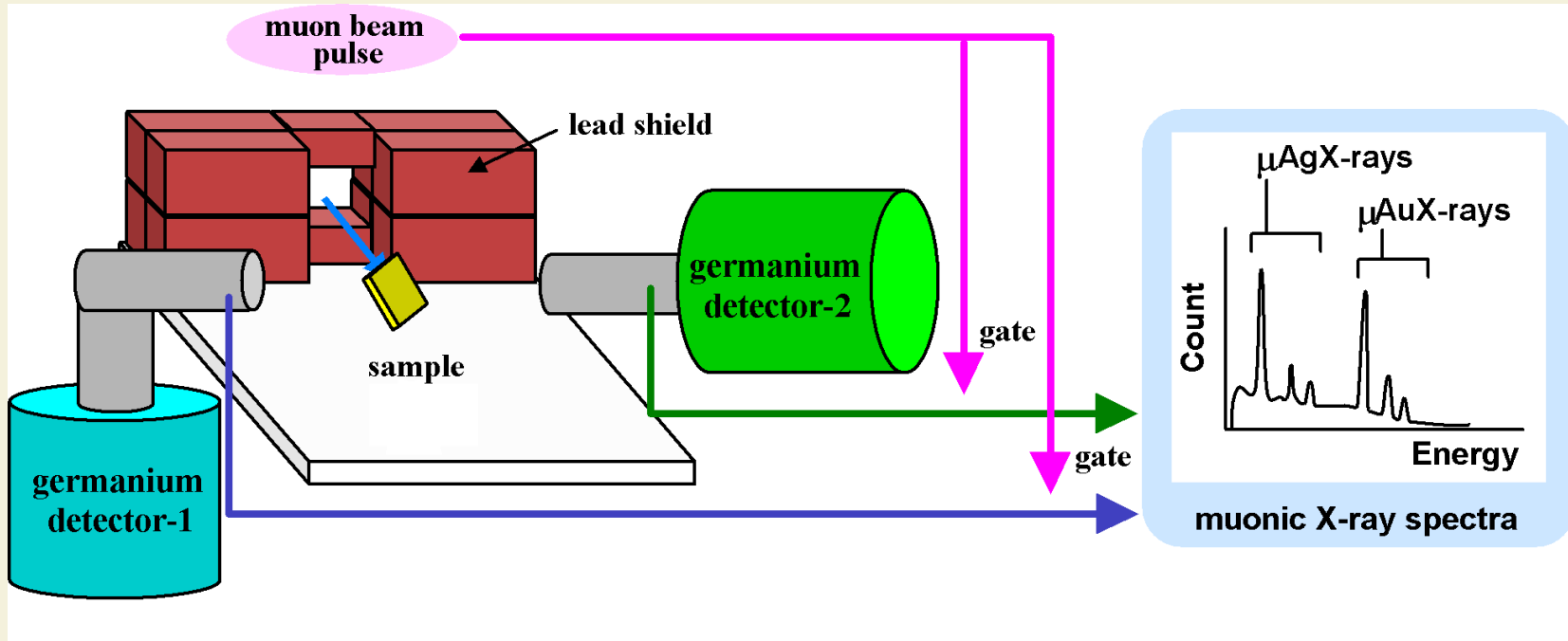
# Experimental

## J-PARC muon facility

Experiments were performed at D2-area in J-PARC/MUSE, Tokai, Ibaraki, Japan

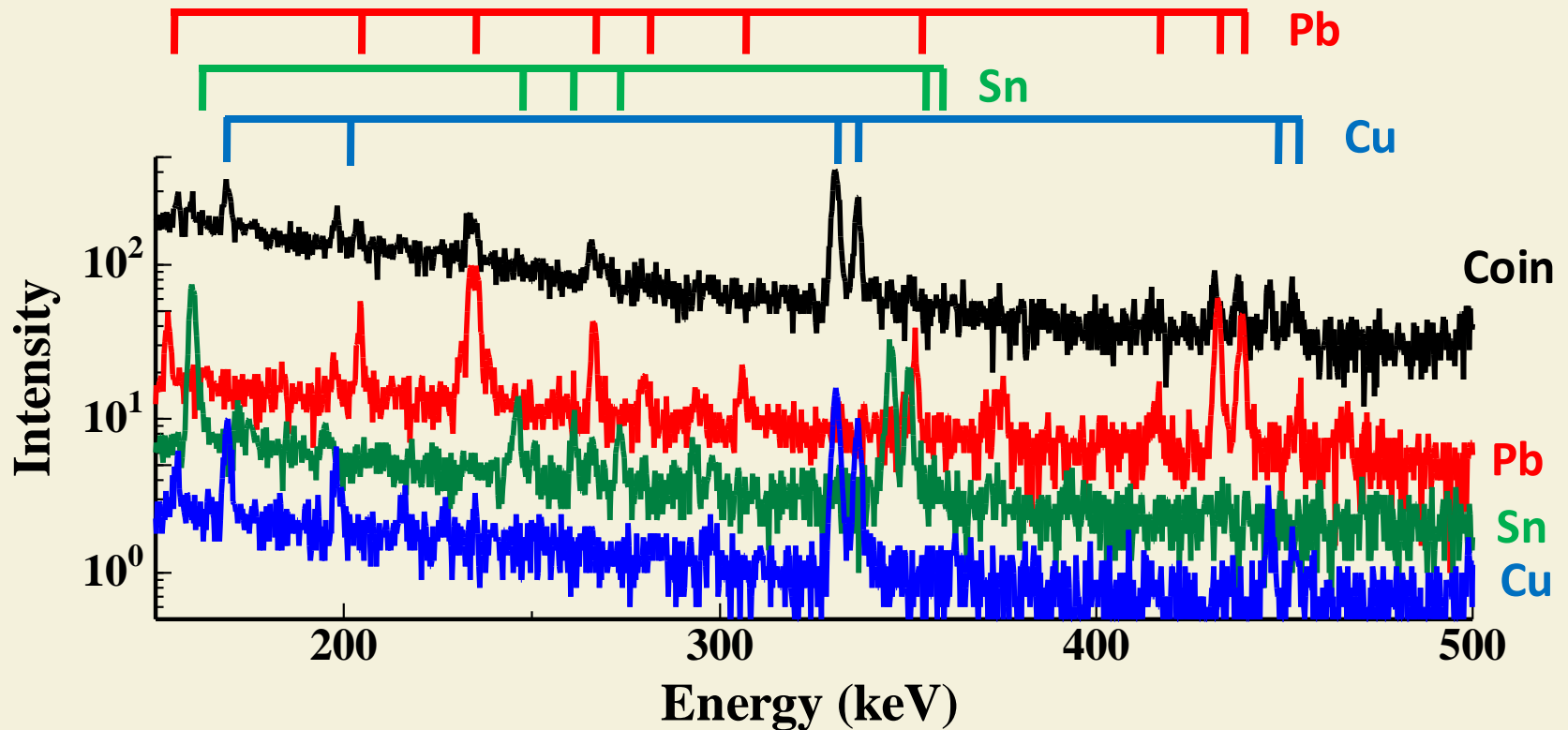


# Experimental setup



**Experimental setup at J-PARC/MUSE D2-ares in 2009**

# Chinese bronze coin

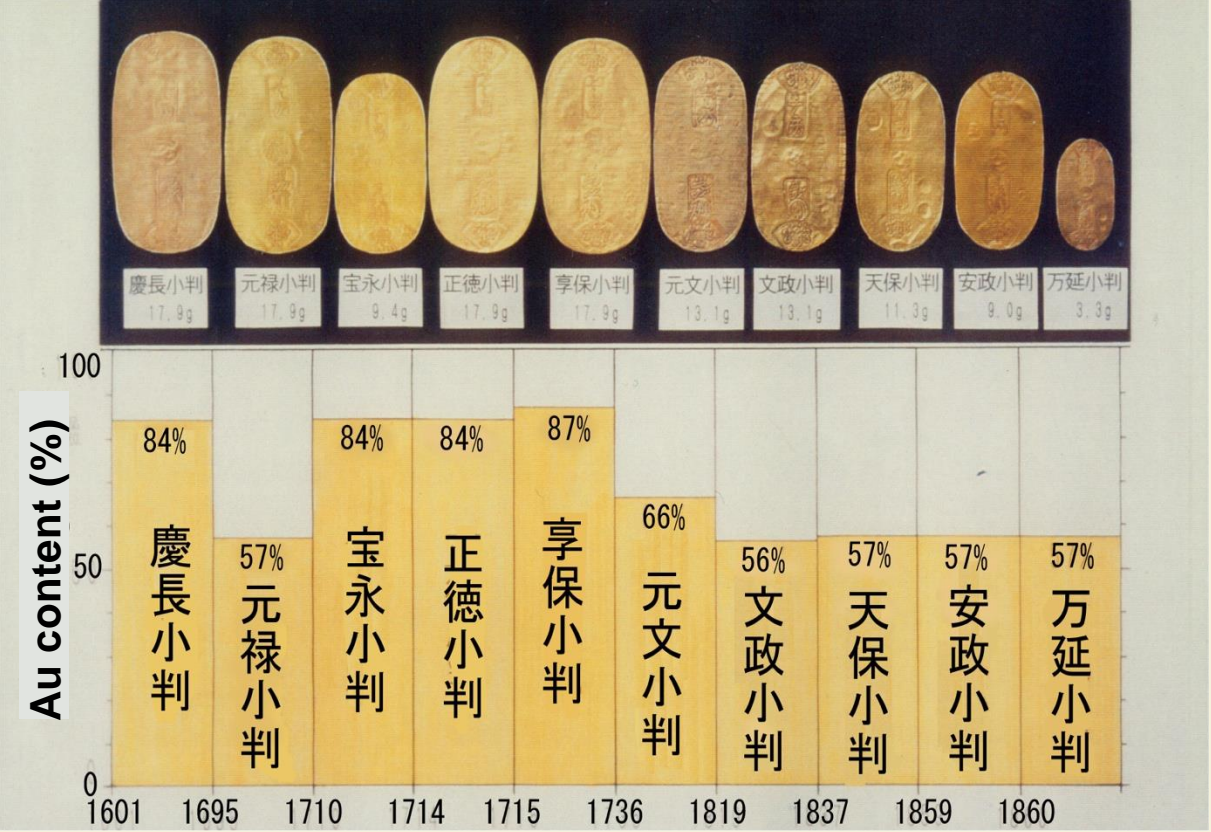


Muonic X-ray spectra of a Chinese Bronze Coin (55000s) and Pb, Sn and Cu metal.

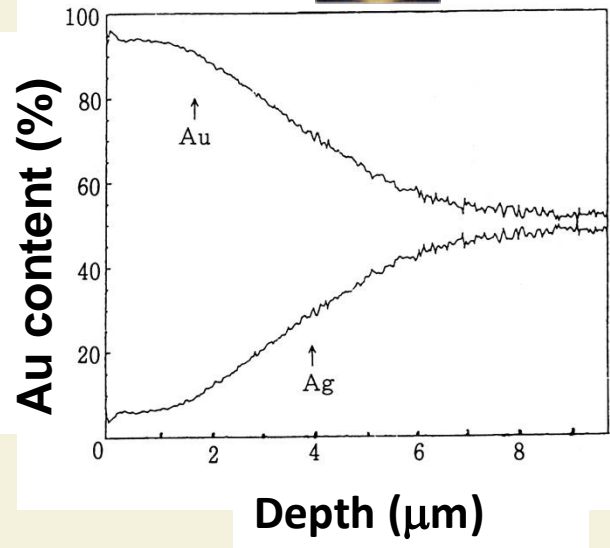
**Sensitivity (in weight) is almost same for all elements.**



# Japanese Gold Coin (Edo Period, 1601 - 1867 )



Tempo Koban



Tempo Koban

Auger Electron Spectroscopy

# Chemical surface treatment of gold coin

Iroage (色揚げ, color dressing)

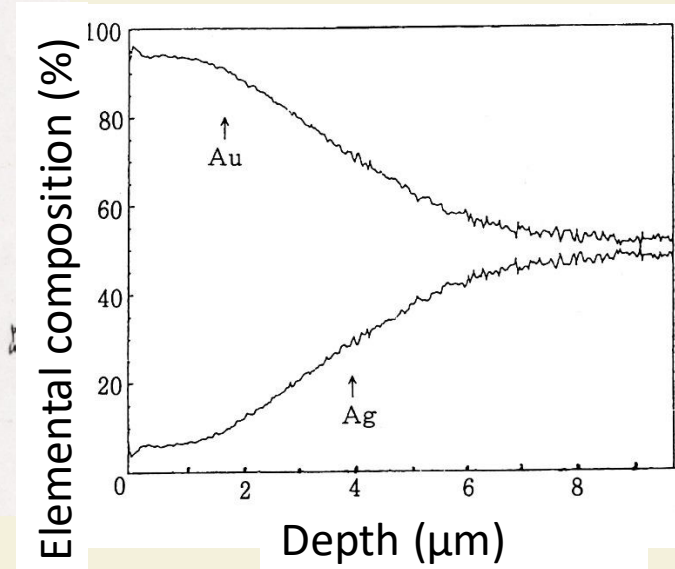
Elemental composition differs depending on depth



色揚げ前(右)と後(左)の小判(複製)

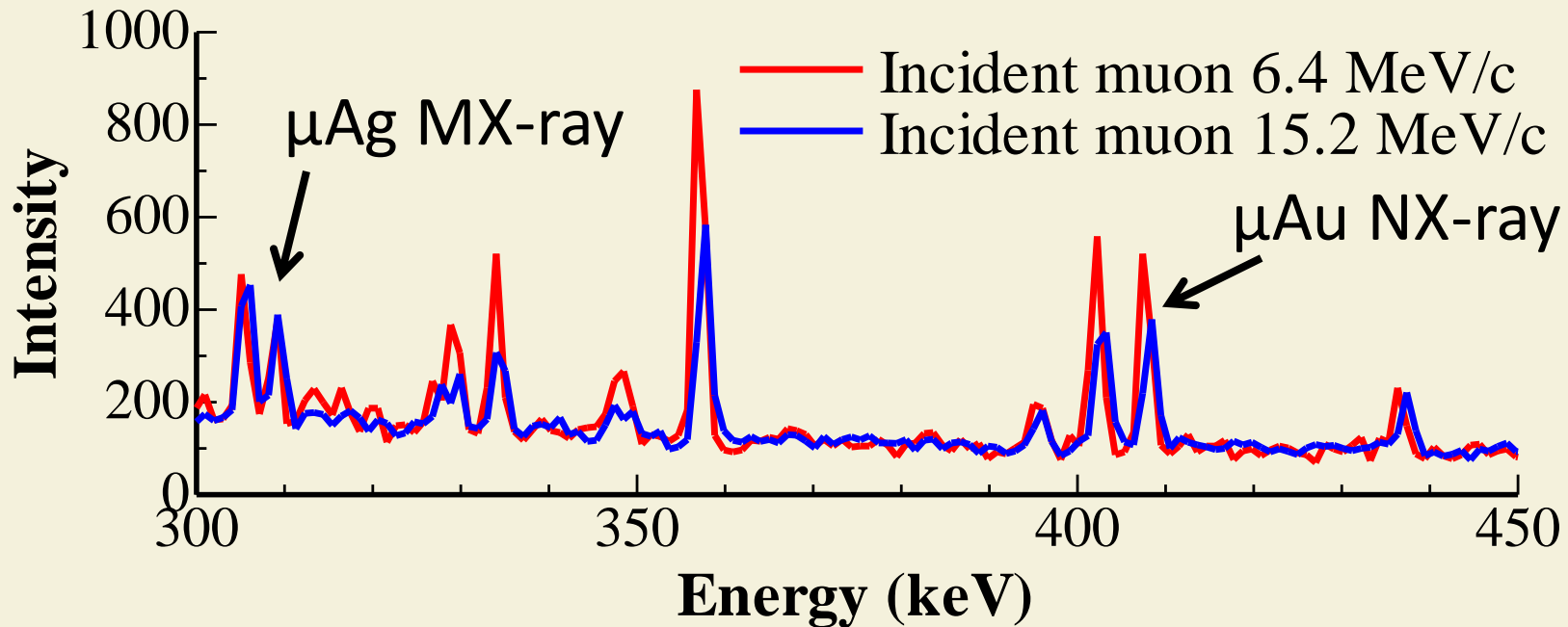


小判の製造工程—色揚げ (小判所絵図)



# Gold coin analysis by muonic X-ray

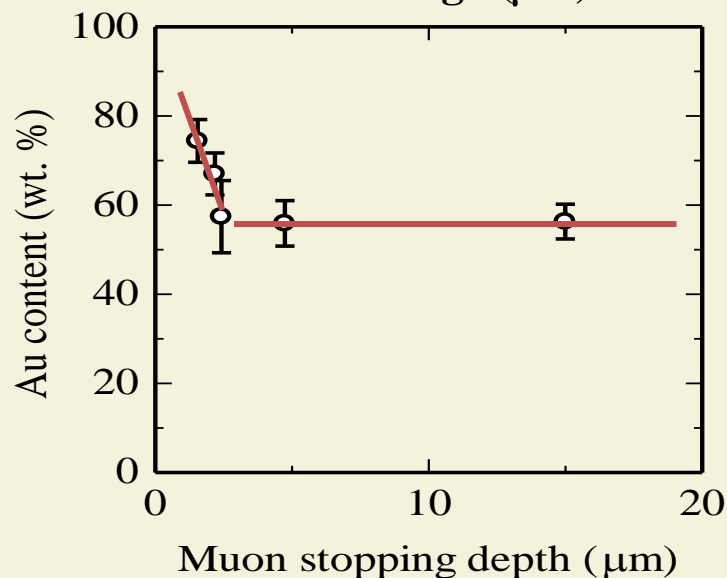
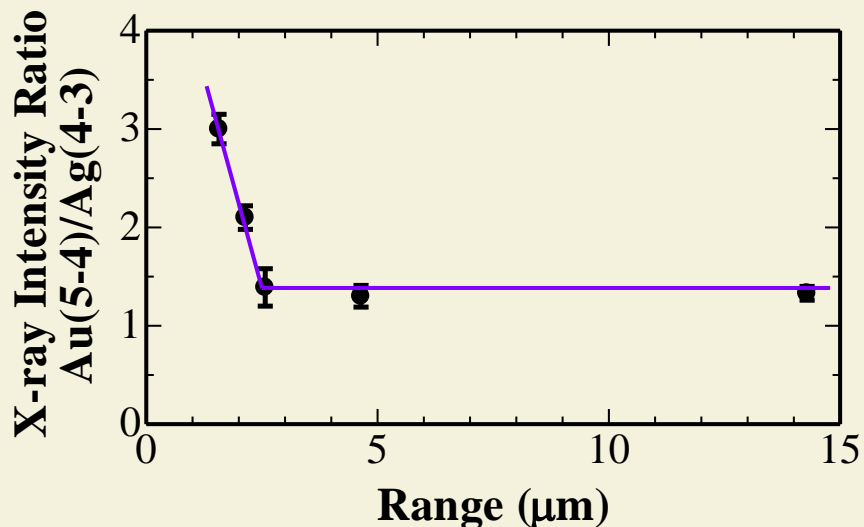
Muonic X-ray spectra for gold coin with different incident muon momenta



Different  $\mu\text{Au}/\mu\text{Ag}$  X-ray intensity ratio was obtained

➔ **Elemental composition varies with depth**

# Gold coin analysis by muonic X-ray



Muon momentum (MeV/c)	Range in gold ( $\mu\text{m}$ )	Mass % of gold
6.4	2.2	$74.4 \pm 4.8$
7.5	3.0	$67.0 \pm 4.7$
8.0	3.6	$57.4 \pm 8.1$
10.2	6.5	$55.9 \pm 5.1$
15.2	20	$56.3 \pm 3.9$

# Depth selective analysis of bronze artefact

Originally brilliant metal of Cu, Sn, Pb

Covered with rust when excavated

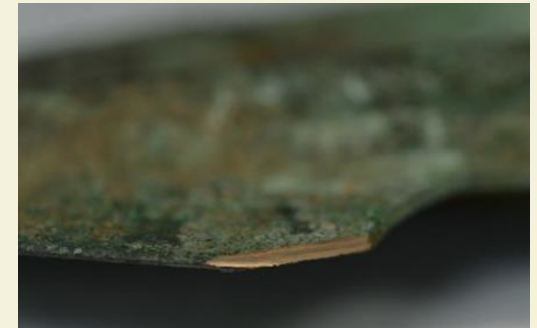
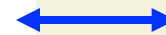


original  
(reconstruction)

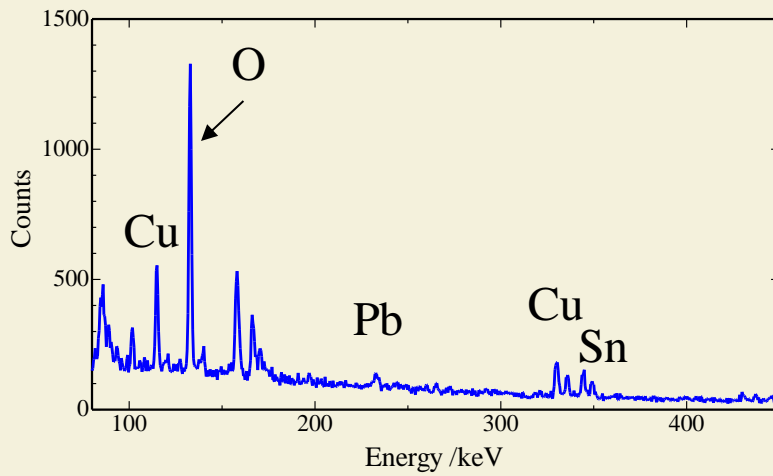
from webpage of Rekihaku



rusty

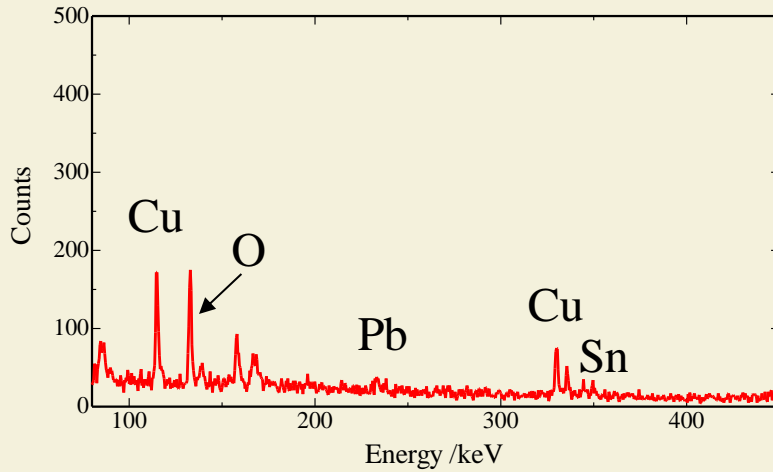


rust covers metallic layer



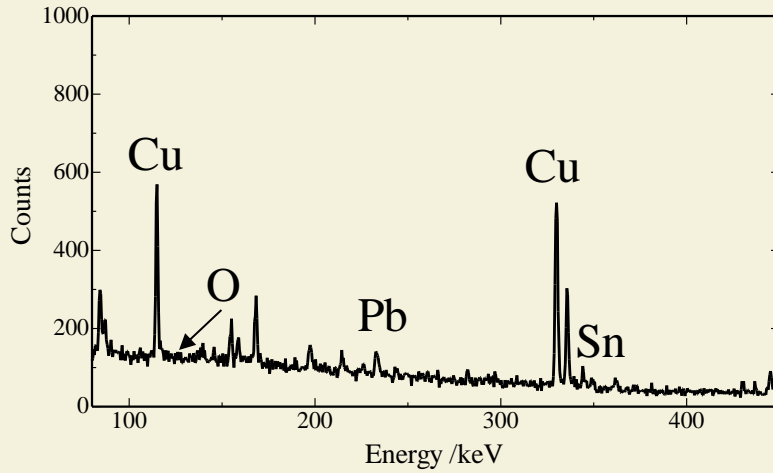
Abundant O  
Low Cu → Rust

depth 0.04 mm  
( 17 MeV/c)



Abundant O  
Low Cu → Rust

depth 0.04 mm  
( 22 MeV/c)



No Oxygen  
High Cu → metal layer

depth 0.04 mm  
( 40 MeV/c)

# Depth selective analysis of bronze artefact

Elemental composition (%)

Depth(mm)	Cu	Sn	Pb	O
0.04 (rust)	24.4	10.4	4.9	60.3
0.10 (rust)	49.6	8.1	6.6	35.7
0.69 (metal)	<b>83.1</b>	<b>9.1</b>	<b>7.9</b>	<b>ND</b>

Original composition



# Asteroid sample

**Multi-elemental** ▪ **Non-destructive** ▪ **Depth-selective**

Negative muon is the answer

- Wide variety of samples:  
historical, archaeological,,etc.  
e. g. asteroid sample by MUSES-C probe



(Our research plan around 2000)

MUSES-C (renamed as Hayabusa) returned in 2010 with a few hundreds of micrograms of dust samples of the asteroid

## Planetary scientists joined.

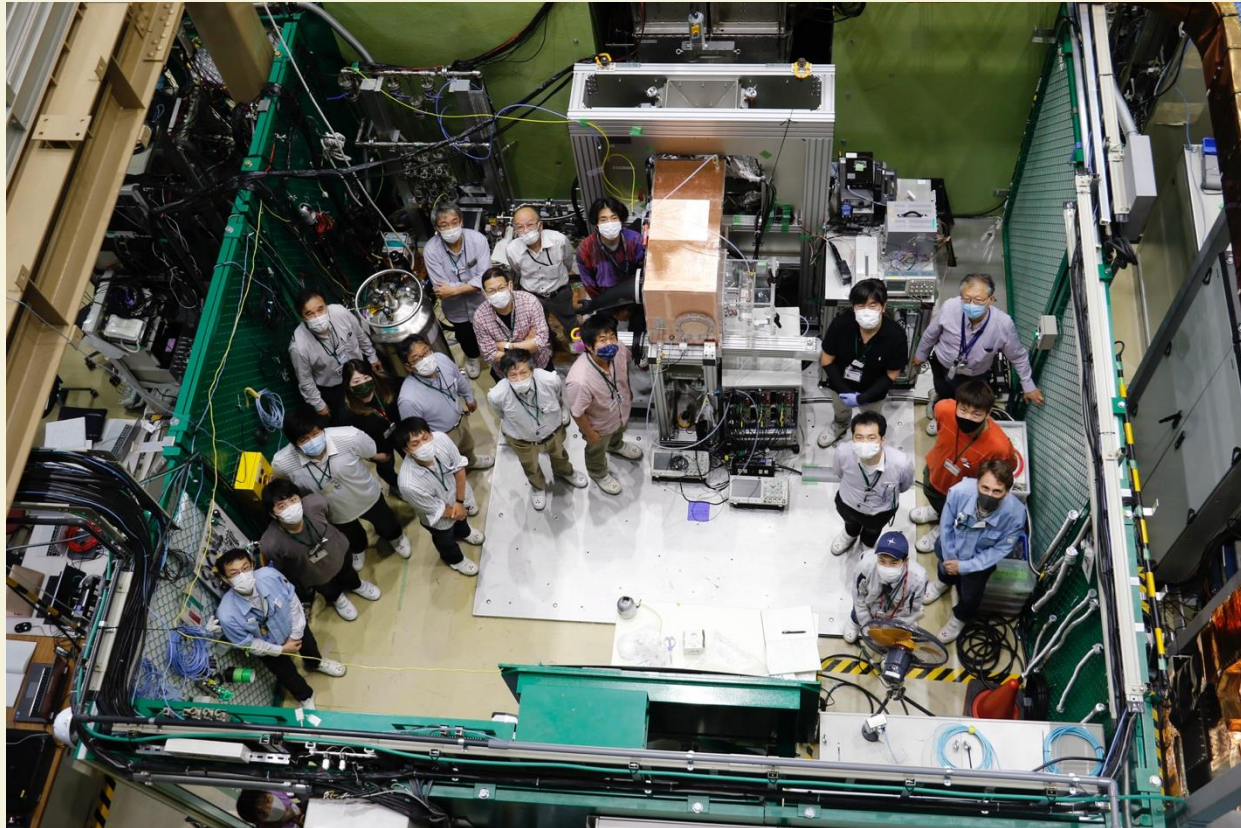
Hayabusa-2 space craft was traveling to asteroid 1999JC3(Ryugu) supposed to have abundant of B, C, N, O, and to return with samples in 2020.



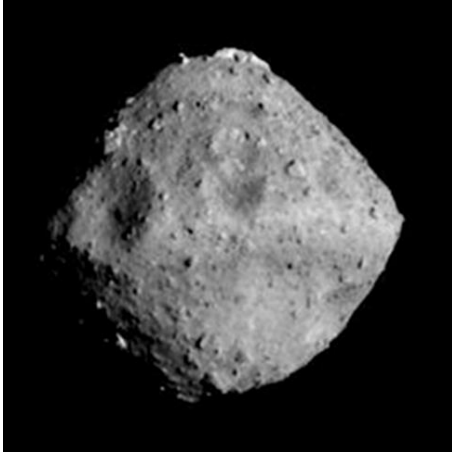
# Asteroid Ryugu Sample Returned by Hayabusa-2

Non-destructive Carbon, Oxygen, Nitrogen analysis

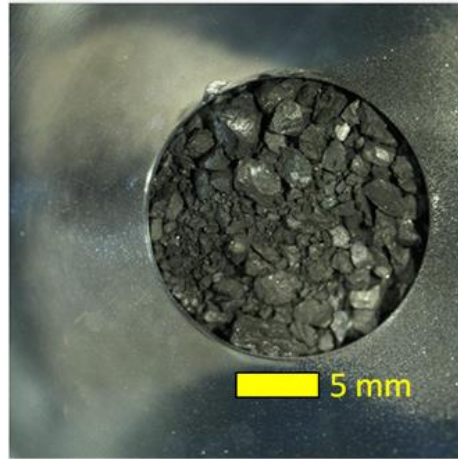
June - July, 2021



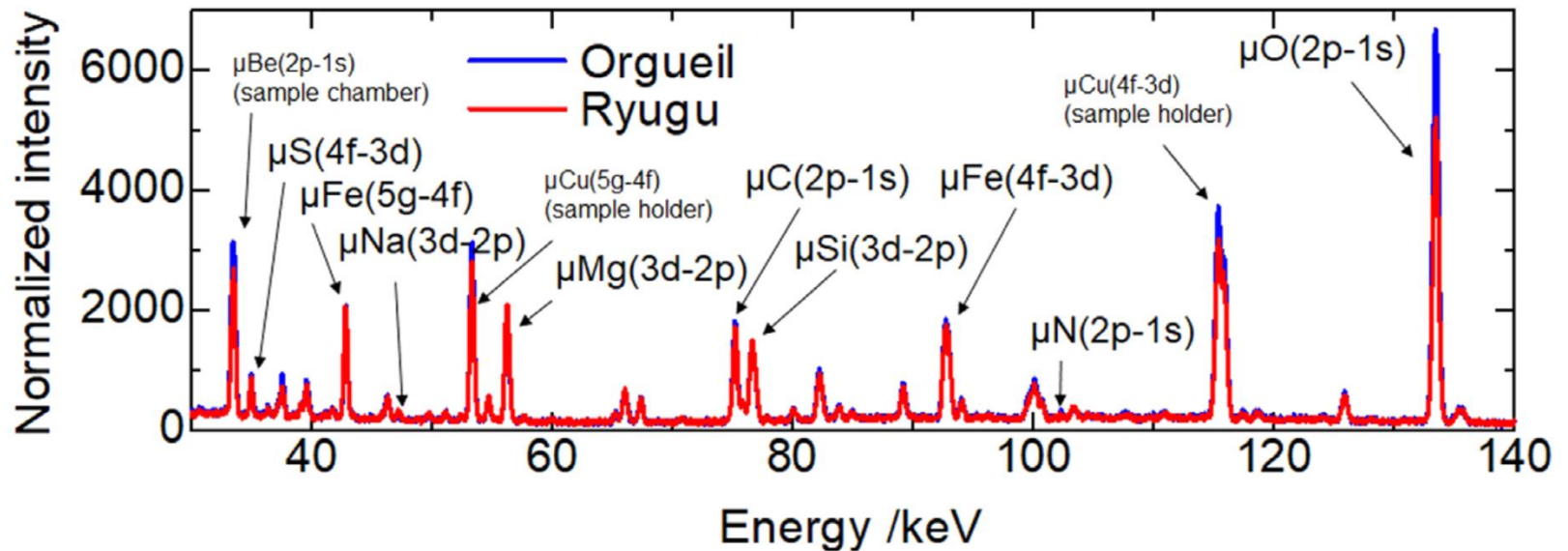
# Asteroid sample



JAXA

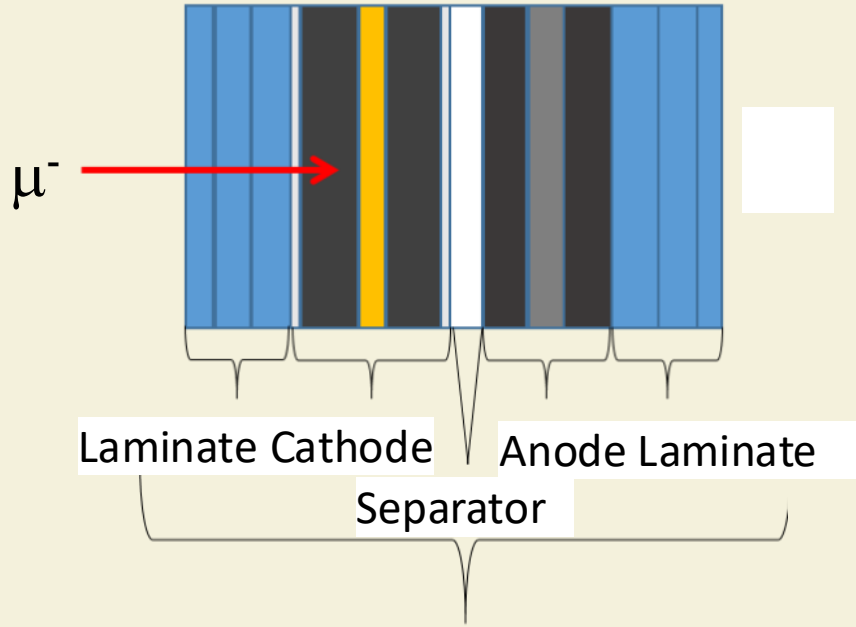


JAXA

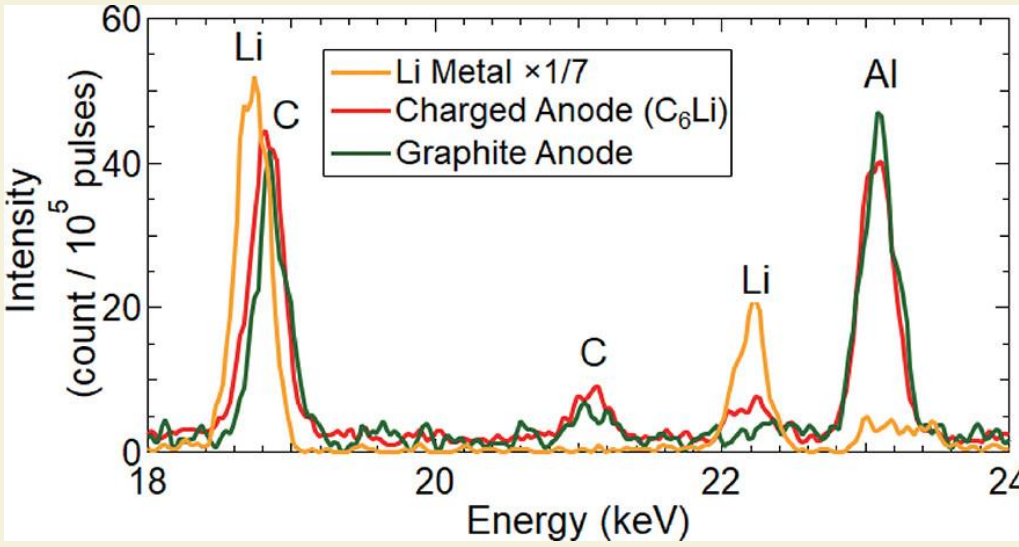


<https://www.science.org/doi/10.1126/science.abn8671>

# Lithium ion battery



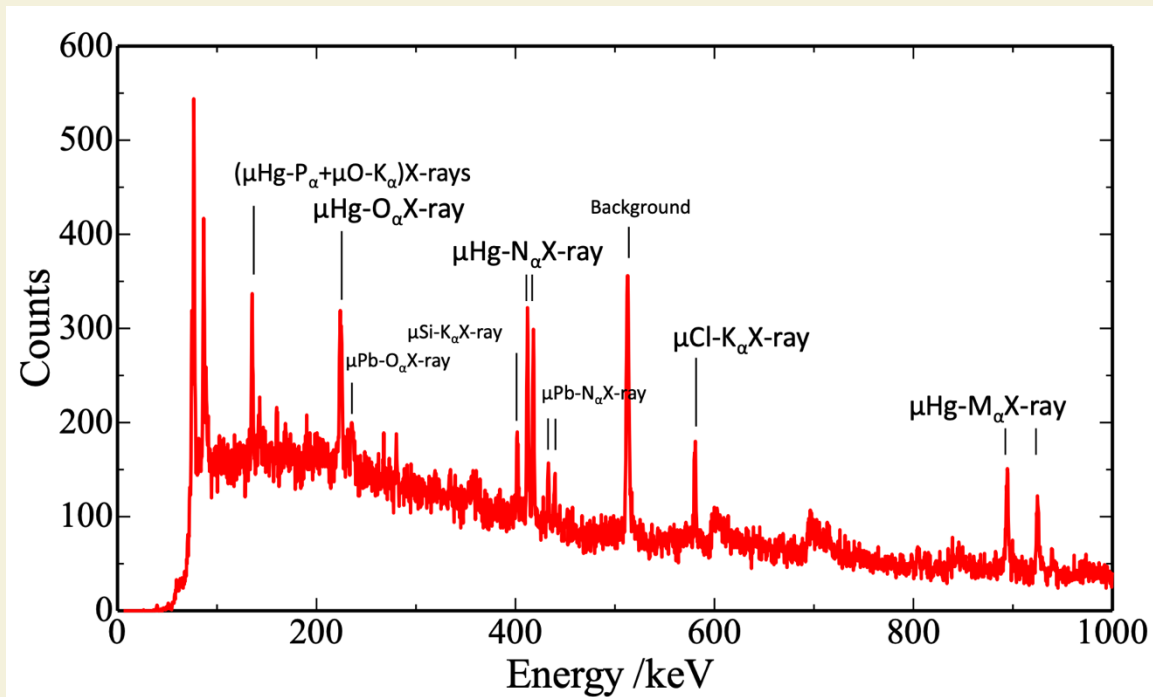
Model battery



*Anal. Chem.* 92 (2020) 194-8200

- cathode ••• graphite (C) 、 coll. electrode (Cu)
- alumina coating ( $Al_2O_3$ )
- anode ••• active material  $Li(Ni_{0.85}Co_{0.15})O_2$
- coll. electrode (Al)
- separator ••• organic (C, H)
- laminate ••• Al, nylon (C, H)

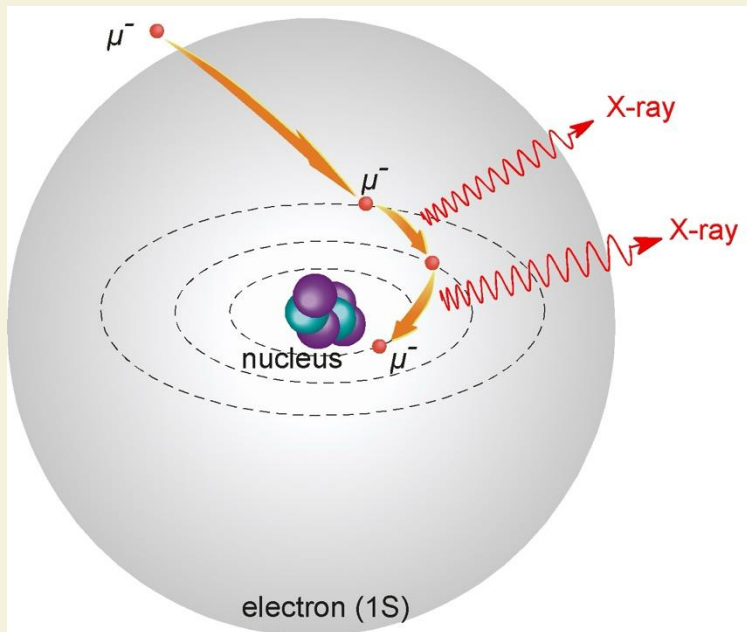
# Unopenable Glass Bottle of Dr. OGATA Koan (1810 – 1863)



# Isotope analysis

$m$ : a reduced mass containing the contribution from the atomic nucleus

Muonic X-ray shows isotope shifts



$$E_n = -\frac{Z^2 m e^4}{8n^2 \epsilon_0^2 h^2}$$

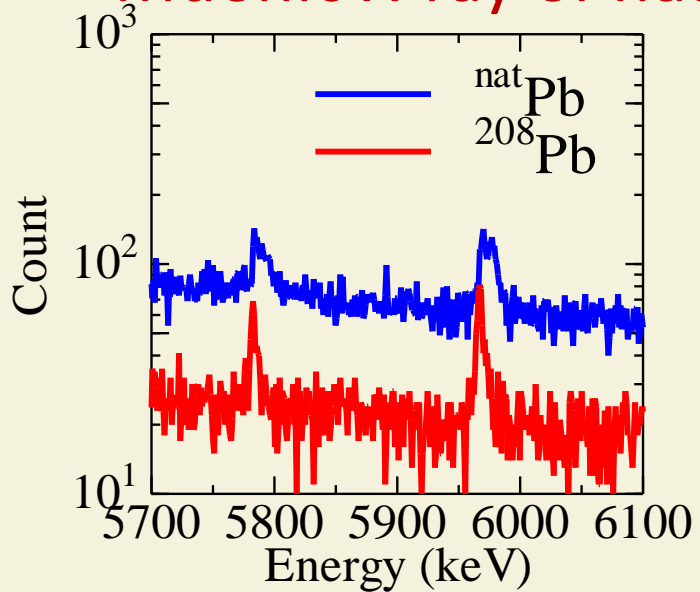
$$r_n = -\frac{4\pi\epsilon_0 n^2 \hbar^2}{Z m e^2}$$

$$\frac{m_\mu}{m_e} \approx 207 \approx \frac{E_\mu}{E_e} \approx \frac{r_e}{r_\mu}$$

**Bohr model**

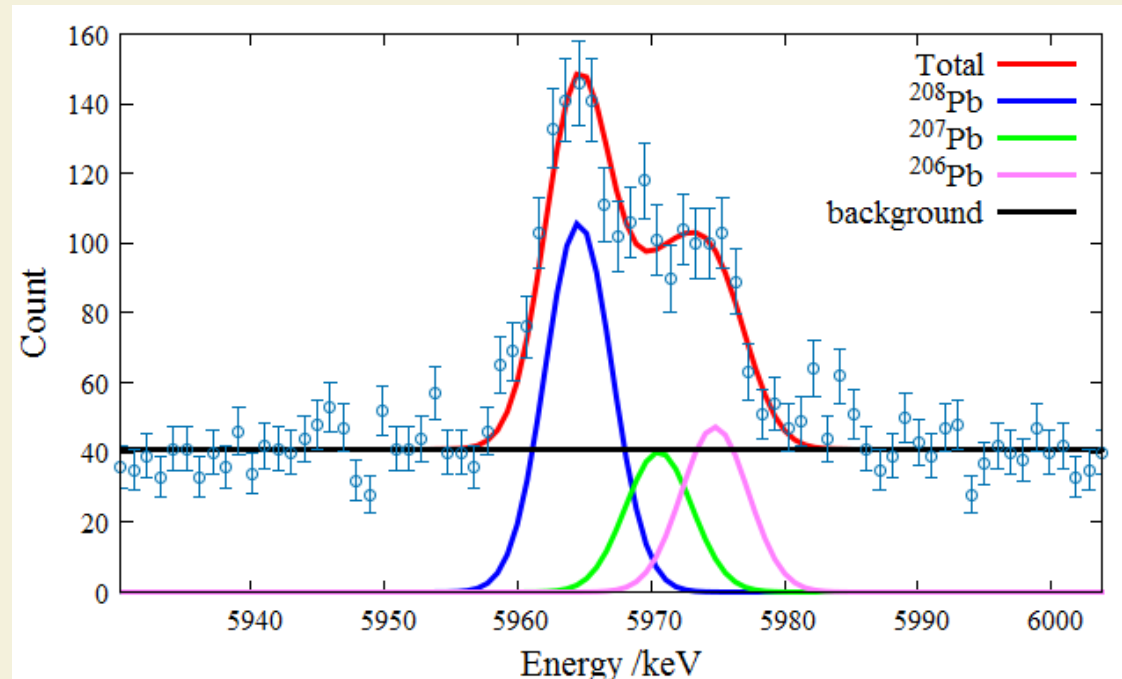
# Isotope analysis

## Muonic X-ray of natural Pb and $^{208}\text{Pb}$



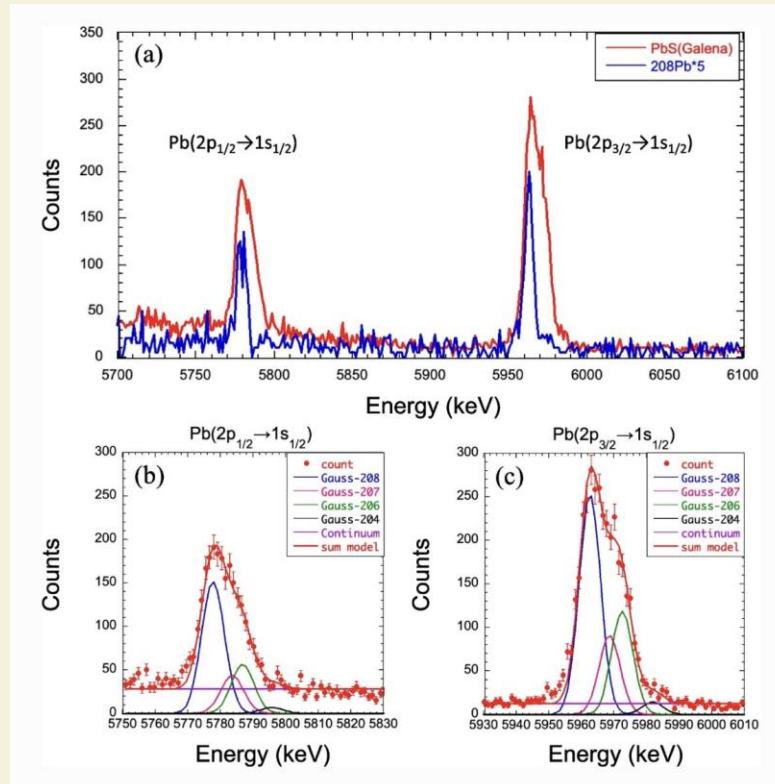
Isotope	abundance(%)
$^{204}\text{Pb}$	1.4(6)
$^{206}\text{Pb}$	24.1(30)
$^{207}\text{Pb}$	22.1(50)
$^{208}\text{Pb}$	52.4(70)

## Decomposition of Muonic $^{nat}\text{Pb}$ X-ray



# Isotope analysis

Galena: PbS



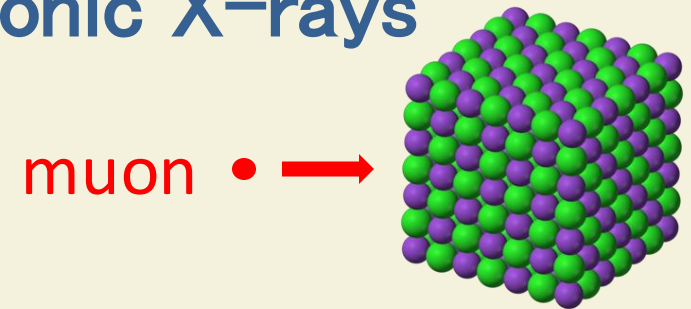
	Average of Muon analysis		LA-ICP mass	
$^{207}\text{Pb}/^{208}\text{Pb}$	0.343	$\pm 0.040$	0.396	$\pm 0.005$
$^{206}\text{Pb}/^{208}\text{Pb}$	0.487	$\pm 0.041$	0.472	$\pm 0.007$
$^{204}\text{Pb}/^{208}\text{Pb}$	0.076	$\pm 0.018$	0.026	$\pm 0.004$

Measured with a DC muon beam at Research Center for Nuclear Physics, Osaka University

# Muon Lifetime Method



# Elemental Analysis with Muonic X-rays



Atomic capture rate is roughly proportional to the atomic number ( $Z$ ) of the muonium capturing atom  
→ muonic X-ray yield is proportional to the weight content of the element

**Components with more than 1% are easily quantified.**

For analysis of minor components...

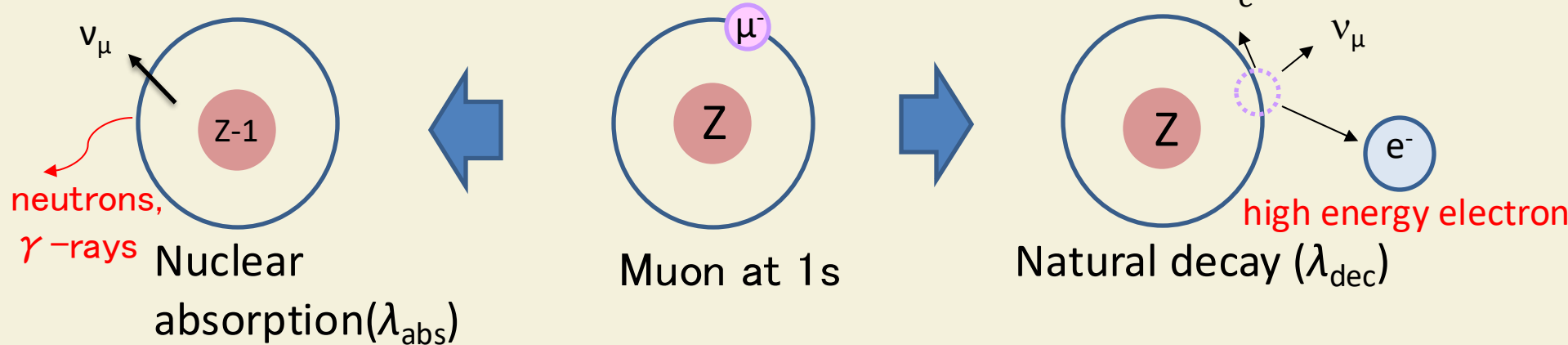
longer beam time for good statistics

accurate evaluation of small peaks

# Muon Lifetime Method

## Negative muon at 1s orbital

no electron emission



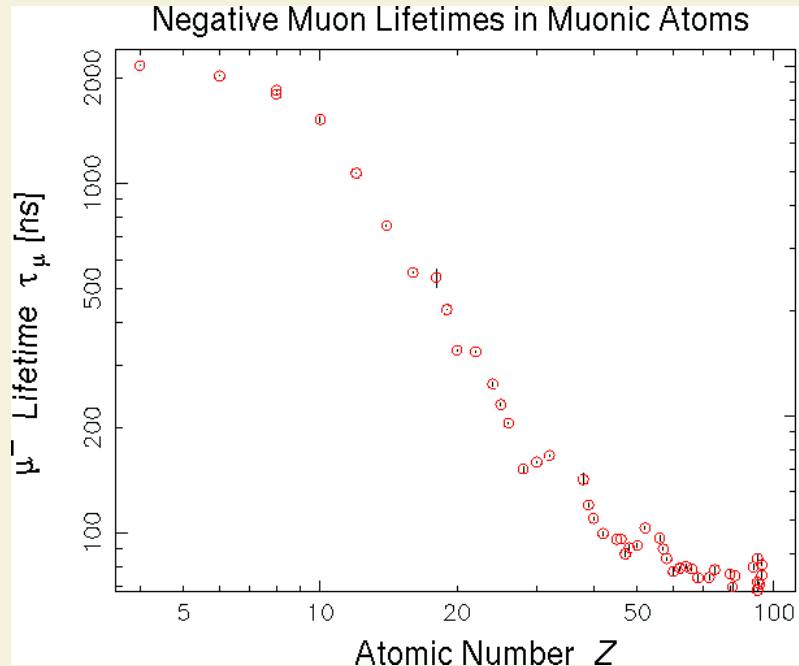
Addition of the nuclear absorption process shortens apparent muon lifetime

$$\tau = \frac{1}{\lambda_{\text{dec}} + \lambda_{\text{abs}}} < \frac{1}{\lambda_{\text{dec}}} (= 2.2 \mu\text{s})$$

Muons in high  $Z$  atoms exhibits shorter apparent lifetime  
→ distinguishable from a long-lived component

# lifetime of negative muon in matter

Element	Mean-life (ns)
$\mu^+$	2197.03 (4)
$^1\text{H}$	2194.90 (7)
$^2\text{H}$	2194.53 (11)
$^3\text{He}$	2186.70 (10)
$^4\text{He}$	2195.31 (5)
$^6\text{Li}$	2175.3 (4)
$^7\text{Li}$	2186.8 (4)
$^9\text{Be}$	2168 (3)
$^{10}\text{B}$	2072 (3)
$^{11}\text{B}$ (lhfs)	2089 (3)
$^{12}\text{C}$	2028 (2)
$^{13}\text{C}$	2037 (8)
$^{14}\text{N}$	1919 (15)
$^{16}\text{O}$	1796 (3)
$^{18}\text{O}$	1844 (5)
$^{19}\text{F}$ (lhfs)	1463 (5)
$^{27}\text{Al}$ (lhfs)	864 (2)
$^{28}\text{Si}$	758 (2)
Ca	334 (2)
Zr	110.4 (10)
Pb	74.8 (4)
Bi	73.4 (4)
Th	77.3 (3)
U	77.0 (4)



## Muon lifetime / $\mu\text{s}$

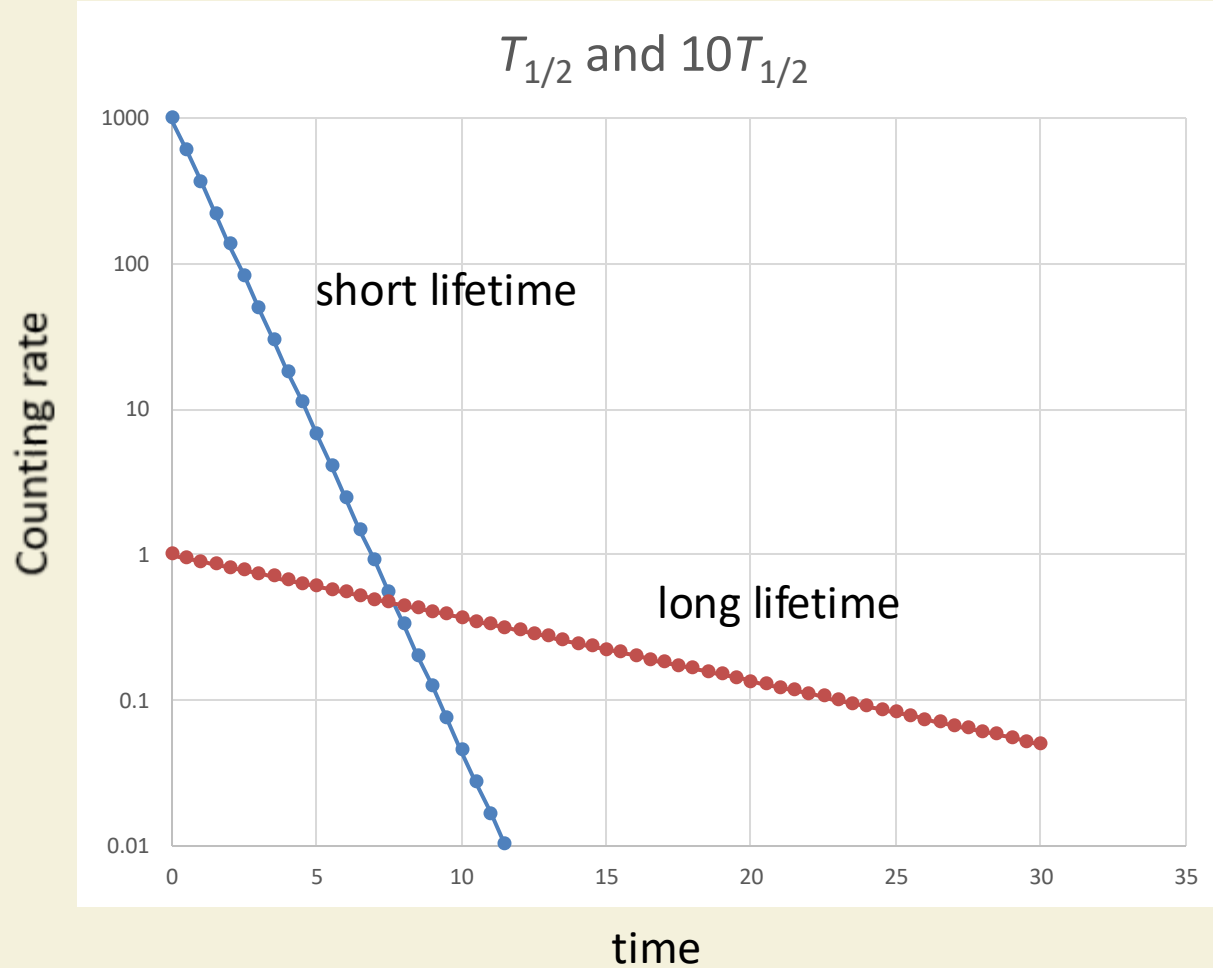
in vacuo 2.197

**Carbon 2.03**

**Iron 0.206**

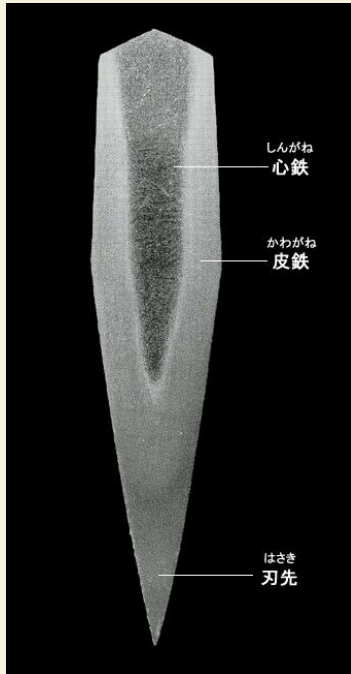
# Lifetime method

Major component with a short half-life and long-lived minor component



Muon lifetime / $\mu\text{s}$	
in vacuo	2.197
Carbon	2.03
Iron	0.206

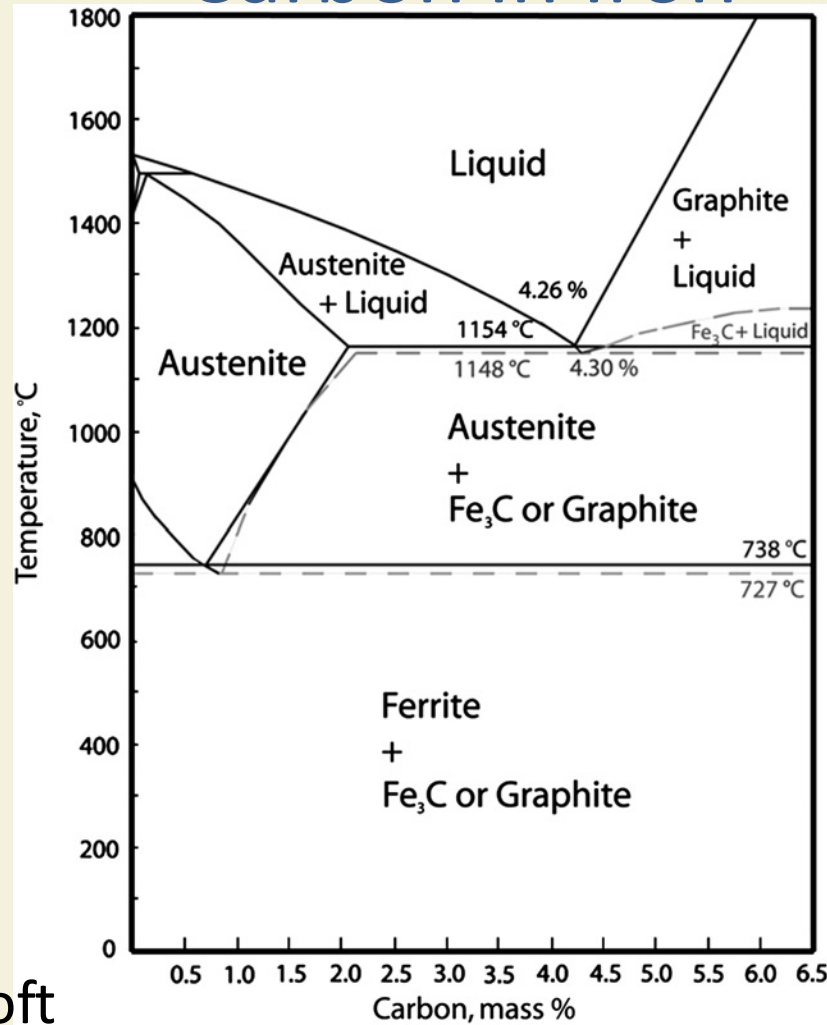
# Japanese sword



## Cross section of a Japanese sword

from Kyoto National Museum web page

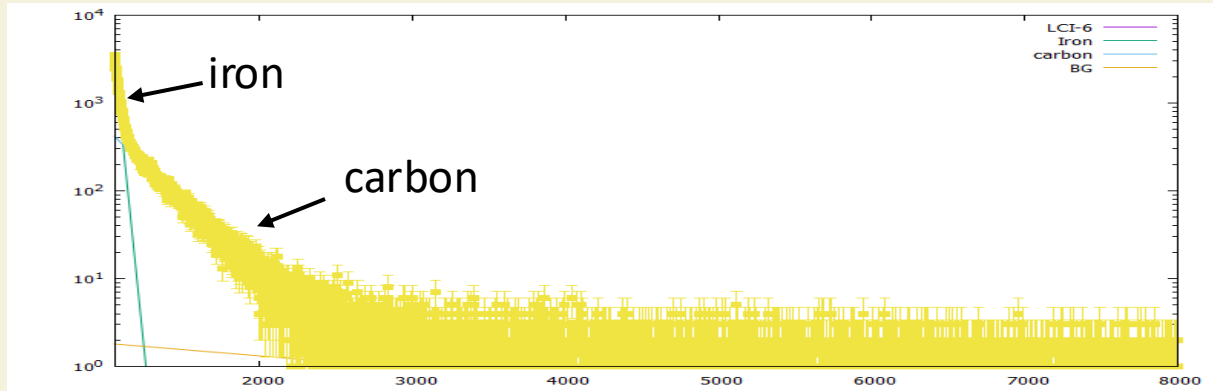
# Carbon in Iron



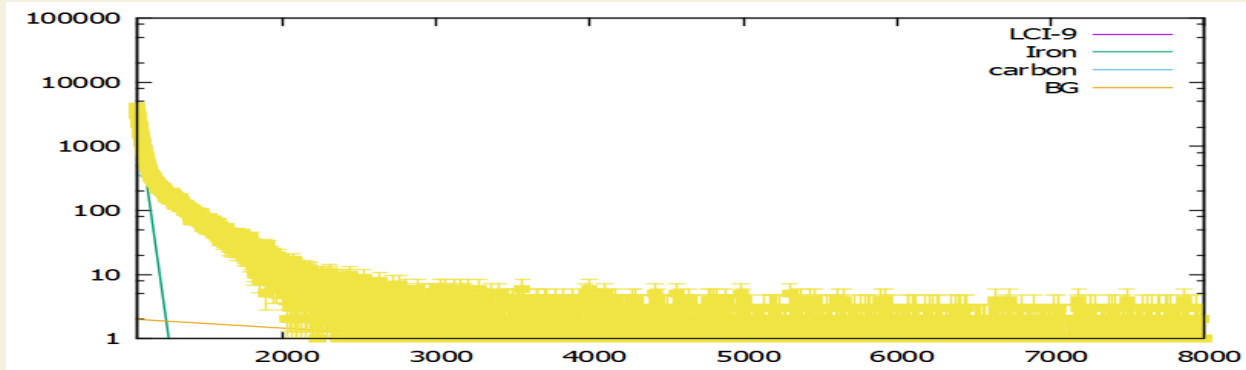
C. N. McCowan et al., Mat. Character., 62 (2011) 807

# Standard iron samples

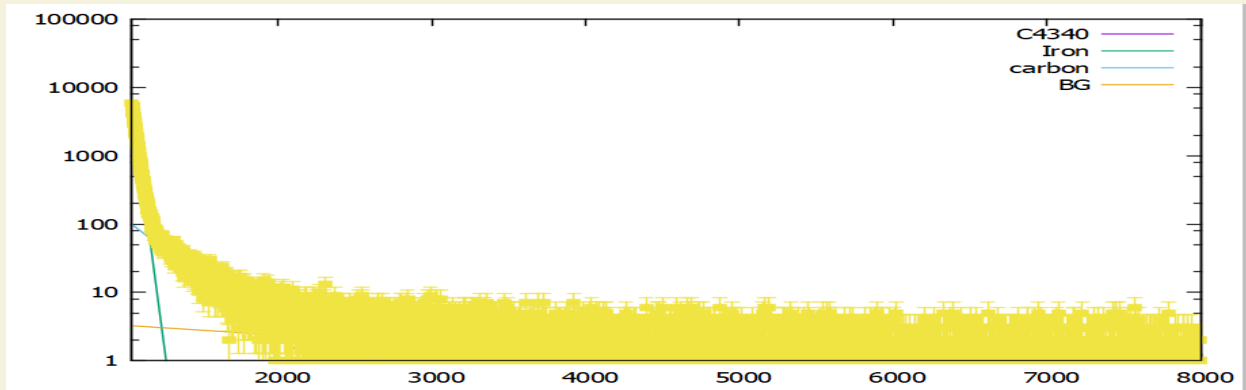
carbon  
4.46%



carbon  
3.22%



carbon  
0.42%



# Depth selective analysis

Stacked sample of iron plates (0.5 mm) with different carbon content

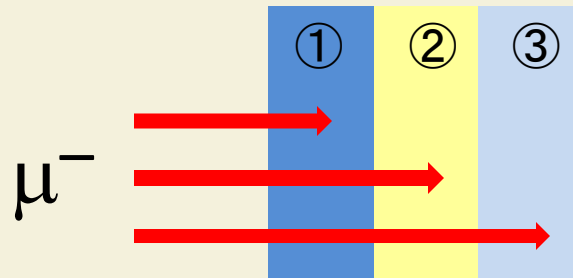
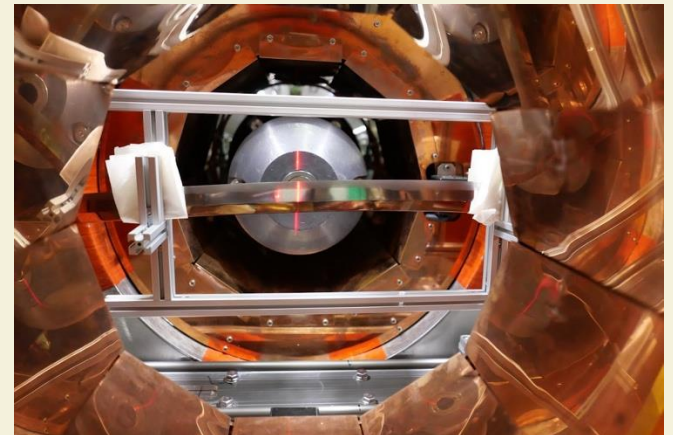


plate	momentum	chemical analysis	muon
①	30 MeV/c	1.03%	1.09(2)%
②	37 MeV/c	0.20%	0.19(1)%
③	43 MeV/c	0.51%	0.50(1)%

# Japanese Sword

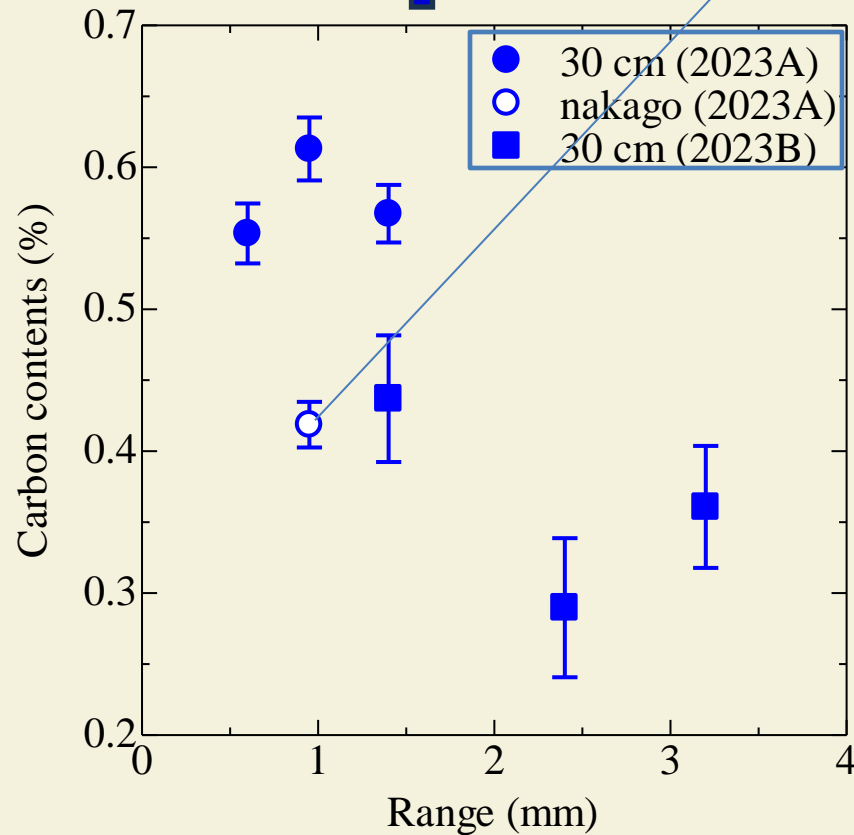


length 51.8 cm  
made in around 16c

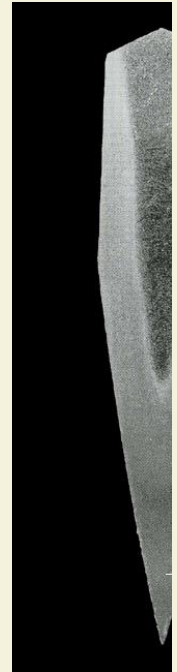




# Japanese Sword



beam size  
5 x 20 mm<sup>2</sup>



# Summary

- Muonic X-ray is a powerful and useful tool for non-destructive, multi-elemental and depth selective elemental analysis.
- Light elements and isotope analysis are possible.
- Uncertainty in depth is about 10% (+- 5%).
- Lifetime method is an alternative for a minor component of a binary element system.

J-PARC MUSE is now accepting proposals from non-science researchers.

ISIS(UK) and PSI(Swiss) group are enjoying Muonic X-ray.

FUME is planned as one of the satellite meetings.